

# 2,5-Cyclohexadiene-1,4-dione, 2-phenyl-

<b>Other names:</b>	p-Benzoquinone, 2-phenyl- Phenyl-p-benzoquinone Phenyl-1,4-benzoquinone Phenylquinone 2-Phenyl-p-benzoquinone 2-Phenyl-1,4-benzoquinone 2-Phenylbenzoquinone Phenylbenzoquinone p-Benzoquinone, phenyl- 2-Phenylquinone NSC 2806 2-Phenylbenzo-1,4-quinone
<b>Inchi:</b>	InChI=1S/C12H8O2/c13-10-6-7-12(14)11(8-10)9-4-2-1-3-5-9/h1-8H
<b>InchiKey:</b>	RLQZIECDMISZHS-UHFFFAOYSA-N
<b>Formula:</b>	C12H8O2
<b>SMILES:</b>	O=C1C=CC(=O)C(c2ccccc2)=C1
<b>Mol. weight [g/mol]:</b>	184.19
<b>CAS:</b>	363-03-1

## Physical Properties

Property code	Value	Unit	Source
ea	2.04 ± 0.06	eV	NIST Webbook
gf	-0.16	kJ/mol	Joback Method
hf	-151.13	kJ/mol	Joback Method
hfus	12.72	kJ/mol	Joback Method
hvap	55.06	kJ/mol	Joback Method
log10ws	-2.28		Crippen Method
logp	1.778		Crippen Method
mcvol	139.860	ml/mol	McGowan Method
pc	3568.53	kPa	Joback Method
tb	663.80	K	Joback Method
tc	939.01	K	Joback Method
tf	413.52	K	Joback Method
vc	0.519	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	348.34	J/mol×K	663.80	Joback Method
cpg	363.99	J/mol×K	709.67	Joback Method
cpg	378.31	J/mol×K	755.54	Joback Method
cpg	391.26	J/mol×K	801.41	Joback Method
cpg	402.82	J/mol×K	847.27	Joback Method
cpg	412.96	J/mol×K	893.14	Joback Method
cpg	421.66	J/mol×K	939.01	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C363031&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C363031&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>ea:</b>	Electron affinity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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