

# Dimethylmalonic acid, 2-naphthylmethyl nonyl ester

<b>Inchi:</b>	InChI=1S/C25H34O4/c1-4-5-6-7-8-9-12-17-28-23(26)25(2,3)24(27)29-19-20-15-16-21-13
<b>InchiKey:</b>	NABPRQIUPHGRX-UHFFFAOYSA-N
<b>Formula:</b>	C25H34O4
<b>SMILES:</b>	CCCCCCCCCOC(=O)C(C)(C)C(=O)OCc1ccc2ccccc2c1
<b>Mol. weight [g/mol]:</b>	398.54

## Physical Properties

Property code	Value	Unit	Source
gf	-95.95	kJ/mol	Joback Method
hf	-641.55	kJ/mol	Joback Method
hfus	49.34	kJ/mol	Joback Method
hvap	92.84	kJ/mol	Joback Method
log10ws	-7.50		Crippen Method
logp	6.203		Crippen Method
mvol	334.770	ml/mol	McGowan Method
pc	1152.22	kPa	Joback Method
rinpol	2934.00		NIST Webbook
rinpol	2934.00		NIST Webbook
tb	971.39	K	Joback Method
tc	1193.44	K	Joback Method
tf	589.89	K	Joback Method
vc	1.286	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1100.96	J/molxK	971.39	Joback Method
cpg	1116.74	J/molxK	1008.40	Joback Method
cpg	1131.41	J/molxK	1045.41	Joback Method
cpg	1145.07	J/molxK	1082.41	Joback Method
cpg	1157.80	J/molxK	1119.42	Joback Method
cpg	1169.72	J/molxK	1156.43	Joback Method
cpg	1180.90	J/molxK	1193.44	Joback Method
dvisc	0.0003833	Paxs	589.89	Joback Method

dvisc	0.0002215	Paxs	653.47	Joback Method
dvisc	0.0001410	Paxs	717.06	Joback Method
dvisc	0.0000967	Paxs	780.64	Joback Method
dvisc	0.0000701	Paxs	844.22	Joback Method
dvisc	0.0000532	Paxs	907.81	Joback Method
dvisc	0.0000419	Paxs	971.39	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U363862&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U363862&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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