

# «gamma»-Calacorene

<b>Inchi:</b>	InChI=1S/C15H20/c1-10(2)13-8-6-12(4)14-7-5-11(3)9-15(13)14/h5-10,12-13H,1-4H3
<b>InchiKey:</b>	LCROQSRTEMTLQQ-UHFFFAOYSA-N
<b>Formula:</b>	C15H20
<b>SMILES:</b>	<chem>Cc1ccc2c(c1)C(C(C)C)C=CC2C</chem>
<b>Mol. weight [g/mol]:</b>	200.32

## Physical Properties

Property code	Value	Unit	Source
gf	237.03	kJ/mol	Joback Method
hf	-40.54	kJ/mol	Joback Method
hfus	22.67	kJ/mol	Joback Method
hvap	52.26	kJ/mol	Joback Method
log10ws	-4.66		Crippen Method
logp	4.408		Crippen Method
mcvol	183.290	ml/mol	McGowan Method
pc	2083.12	kPa	Joback Method
ripol	1553.00		NIST Webbook
ripol	1553.00		NIST Webbook
ripol	1558.00		NIST Webbook
ripol	1550.00		NIST Webbook
ripol	1556.00		NIST Webbook
ripol	1571.00		NIST Webbook
ripol	1984.00		NIST Webbook
ripol	1984.00		NIST Webbook
ripol	1984.00		NIST Webbook
ripol	1984.00		NIST Webbook
ripol	1960.00		NIST Webbook
ripol	1984.00		NIST Webbook
ripol	1984.00		NIST Webbook
ripol	1984.00		NIST Webbook
ripol	1984.00		NIST Webbook
ripol	1984.00		NIST Webbook
ripol	1984.00		NIST Webbook
ripol	1984.00		NIST Webbook
ripol	1984.00		NIST Webbook
tb	584.30	K	Joback Method
tc	804.86	K	Joback Method

tf	306.21	K	Joback Method
vc	0.696	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	462.66	J/mol×K	584.30	Joback Method
cpg	482.45	J/mol×K	621.06	Joback Method
cpg	501.03	J/mol×K	657.82	Joback Method
cpg	518.45	J/mol×K	694.58	Joback Method
cpg	534.78	J/mol×K	731.34	Joback Method
cpg	550.06	J/mol×K	768.10	Joback Method
cpg	564.36	J/mol×K	804.86	Joback Method
dvisc	0.0015832	Paxs	306.21	Joback Method
dvisc	0.0010106	Paxs	352.56	Joback Method
dvisc	0.0007160	Paxs	398.91	Joback Method
dvisc	0.0005450	Paxs	445.25	Joback Method
dvisc	0.0004368	Paxs	491.60	Joback Method
dvisc	0.0003637	Paxs	537.95	Joback Method
dvisc	0.0003117	Paxs	584.30	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R237233&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R237233&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions

<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log10 of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>p<sub>c</sub>:</b>	Critical Pressure
<b>r<sub>inpol</sub>:</b>	Non-polar retention indices
<b>r<sub>ipol</sub>:</b>	Polar retention indices
<b>t<sub>b</sub>:</b>	Normal Boiling Point Temperature
<b>t<sub>c</sub>:</b>	Critical Temperature
<b>t<sub>f</sub>:</b>	Normal melting (fusion) point
<b>v<sub>c</sub>:</b>	Critical Volume

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