

# Phenol, 3-ethyl-

|                             |  |
|-----------------------------|--|
| <b>Other names:</b>         | 1-ETHYL-3-HYDROXYBENZENE<br>1-Hydroxy-3-ethylbenzene<br>3-ETHYLPHENOL<br>Phenol, m-ethyl-<br>m-Ethylphenol |
| <b>Inchi:</b>               | InChI=1S/C8H10O/c1-2-7-4-3-5-8(9)6-7/h3-6,9H,2H2,1H3   |
| <b>InchiKey:</b>            | HMNKTRSOROOSPP-UHFFFAOYSA-N  |
| <b>Formula:</b>             | C8H10O   |
| <b>SMILES:</b>              | CCc1cccc(O)c1  |
| <b>Mol. weight [g/mol]:</b> | 122.16   |
| <b>CAS:</b>                 | 620-17-7   |

## Physical Properties

| Property code | Value           | Unit   | Source         |
|---------------|-----------------|--------|----------------|
| chl           | -4362.90 ± 1.50 | kJ/mol | NIST Webbook   |
| gf            | -25.73          | kJ/mol | Joback Method  |
| hf            | -146.60         | kJ/mol | KDB            |
| hf            | -146.20 ± 1.70  | kJ/mol | NIST Webbook   |
| hfl           | -214.30 ± 1.60  | kJ/mol | NIST Webbook   |
| hfus          | 16.30           | kJ/mol | Joback Method  |
| hvap          | 68.20 ± 0.10    | kJ/mol | NIST Webbook   |
| hvap          | 68.10           | kJ/mol | NIST Webbook   |
| log10ws       | -1.82           |        | Crippen Method |
| logp          | 1.955           |        | Crippen Method |
| mcvol         | 105.690         | ml/mol | McGowan Method |
| pc            | 4409.10         | kPa    | Joback Method  |
| rinpol        | 1143.00         |        | NIST Webbook   |
| rinpol        | 1166.00         |        | NIST Webbook   |
| rinpol        | 1169.00         |        | NIST Webbook   |
| rinpol        | 1142.70         |        | NIST Webbook   |
| rinpol        | 1171.00         |        | NIST Webbook   |
| rinpol        | 1183.70         |        | NIST Webbook   |
| rinpol        | 1175.00         |        | NIST Webbook   |
| rinpol        | 1169.00         |        | NIST Webbook   |
| rinpol        | 1153.00         |        | NIST Webbook   |
| rinpol        | 1167.00         |        | NIST Webbook   |
| rinpol        | 1167.00         |        | NIST Webbook   |

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|--------|---------|--------------|
| rinpol | 1169.00 | NIST Webbook |
| rinpol | 1138.00 | NIST Webbook |
| rinpol | 1195.00 | NIST Webbook |
| rinpol | 1175.00 | NIST Webbook |
| rinpol | 1176.00 | NIST Webbook |
| rinpol | 1135.00 | NIST Webbook |
| rinpol | 1144.00 | NIST Webbook |
| rinpol | 1130.00 | NIST Webbook |
| rinpol | 1172.00 | NIST Webbook |
| rinpol | 1139.00 | NIST Webbook |
| rinpol | 1138.00 | NIST Webbook |
| rinpol | 1170.00 | NIST Webbook |
| rinpol | 1169.00 | NIST Webbook |
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| rinpol | 1176.00 | NIST Webbook |
| rinpol | 1170.00 | NIST Webbook |
| rinpol | 1147.00 | NIST Webbook |
| rinpol | 1160.00 | NIST Webbook |
| rinpol | 197.80  | NIST Webbook |
| rinpol | 190.60  | NIST Webbook |
| rinpol | 194.60  | NIST Webbook |
| rinpol | 196.96  | NIST Webbook |
| rinpol | 196.80  | NIST Webbook |
| rinpol | 196.96  | NIST Webbook |
| rinpol | 196.80  | NIST Webbook |
| rinpol | 1140.00 | NIST Webbook |
| rinpol | 1160.00 | NIST Webbook |
| rinpol | 1183.70 | NIST Webbook |
| rinpol | 1167.00 | NIST Webbook |
| rinpol | 1143.00 | NIST Webbook |
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| rinpol | 1140.00 | NIST Webbook |
| rinpol | 1189.00 | NIST Webbook |
| rinpol | 1189.00 | NIST Webbook |
| ripol  | 2191.00 | NIST Webbook |
| ripol  | 2190.00 | NIST Webbook |
| ripol  | 2195.00 | NIST Webbook |
| ripol  | 2218.00 | NIST Webbook |
| ripol  | 2151.00 | NIST Webbook |
| ripol  | 2141.00 | NIST Webbook |
| ripol  | 2150.00 | NIST Webbook |
| ripol  | 2195.00 | NIST Webbook |
| ripol  | 2189.00 | NIST Webbook |

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|-------|---------------|----------------------|---------------|
| ripol | 2188.00       |                      | NIST Webbook  |
| ripol | 2189.00       |                      | NIST Webbook  |
| ripol | 2190.00       |                      | NIST Webbook  |
| ripol | 2205.00       |                      | NIST Webbook  |
| ripol | 2168.00       |                      | NIST Webbook  |
| ripol | 2169.00       |                      | NIST Webbook  |
| ripol | 2171.00       |                      | NIST Webbook  |
| ripol | 2210.00       |                      | NIST Webbook  |
| ripol | 2170.00       |                      | NIST Webbook  |
| ripol | 2167.00       |                      | NIST Webbook  |
| ripol | 2171.00       |                      | NIST Webbook  |
| ripol | 2194.00       |                      | NIST Webbook  |
| ripol | 2194.00       |                      | NIST Webbook  |
| ripol | 2181.00       |                      | NIST Webbook  |
| ripol | 2170.00       |                      | NIST Webbook  |
| tb    | 491.57        | K                    | KDB           |
| tc    | 718.80        | K                    | KDB           |
| tc    | 716.45 ± 0.15 | K                    | NIST Webbook  |
| tf    | 269.15 ± 2.00 | K                    | NIST Webbook  |
| tf    | 269.00        | K                    | KDB           |
| tf    | 269.15 ± 2.00 | K                    | NIST Webbook  |
| tf    | 269.15 ± 2.00 | K                    | NIST Webbook  |
| tf    | 269.15 ± 2.00 | K                    | NIST Webbook  |
| tf    | 269.15 ± 2.00 | K                    | NIST Webbook  |
| tf    | 268.65 ± 1.00 | K                    | NIST Webbook  |
| tf    | 269.15 ± 2.00 | K                    | NIST Webbook  |
| tf    | 269.15 ± 2.00 | K                    | NIST Webbook  |
| vc    | 0.342         | m <sup>3</sup> /kmol | Joback Method |

## Temperature Dependent Properties

| Property code | Value     | Unit    | Temperature [K] | Source        |
|---------------|-----------|---------|-----------------|---------------|
| cpg           | 284.56    | J/mol×K | 715.75          | Joback Method |
| cpg           | 224.79    | J/mol×K | 489.74          | Joback Method |
| cpg           | 236.73    | J/mol×K | 527.41          | Joback Method |
| cpg           | 247.77    | J/mol×K | 565.08          | Joback Method |
| cpg           | 258.00    | J/mol×K | 602.74          | Joback Method |
| cpg           | 267.49    | J/mol×K | 640.41          | Joback Method |
| cpg           | 276.31    | J/mol×K | 678.08          | Joback Method |
| dvisc         | 0.0000969 | Paxs    | 489.74          | Joback Method |
| dvisc         | 0.0051808 | Paxs    | 318.06          | Joback Method |

|       |           |        |        |               |
|-------|-----------|--------|--------|---------------|
| dvisc | 0.0020303 | Paxs   | 346.67 | Joback Method |
| dvisc | 0.0009178 | Paxs   | 375.29 | Joback Method |
| dvisc | 0.0004643 | Paxs   | 403.90 | Joback Method |
| dvisc | 0.0002571 | Paxs   | 432.51 | Joback Method |
| dvisc | 0.0001531 | Paxs   | 461.13 | Joback Method |
| hvapt | 48.80     | kJ/mol | 417.50 | NIST Webbook  |
| hvapt | 53.10     | kJ/mol | 474.00 | NIST Webbook  |
| hvapt | 68.10     | kJ/mol | 300.00 | NIST Webbook  |
| hvapt | 58.30     | kJ/mol | 417.50 | NIST Webbook  |
| hvapt | 56.50     | kJ/mol | 417.50 | NIST Webbook  |
| hvapt | 55.20     | kJ/mol | 417.50 | NIST Webbook  |
| hvapt | 53.70     | kJ/mol | 417.50 | NIST Webbook  |
| rho1  | 1025.00   | kg/m3  | 273.00 | KDB           |

## Pressure Dependent Properties

| Property code | Value  | Unit | Pressure [kPa] | Source       |
|---------------|--------|------|----------------|--------------|
| tbrp          | 382.20 | K    | 2.00           | NIST Webbook |

## Correlations

| Information                 | Value                         |
|-----------------------------|-------------------------------|
| Property code               | pvap                          |
| Equation                    | $\ln(P_{vp}) = A + B/(T + C)$ |
| Coeff. A                    | 1.56730e+01                   |
| Coeff. B                    | -4.54561e+03                  |
| Coeff. C                    | -7.78060e+01                  |
| Temperature range (K), min. | 373.26                        |
| Temperature range (K), max. | 516.51                        |

| Information   | Value  |
|---------------|--|
| Property code | pvap   |
| Equation      | $\ln(P_{vp}) = A + B/T + C \cdot \ln(T) + D \cdot T^2$ |
| Coeff. A      | 1.13752e+02  |
| Coeff. B      | -1.18918e+04   |
| Coeff. C      | -1.38880e+01   |
| Coeff. D      | 4.72667e-06  |

|                             |        |
|-----------------------------|--------|
| Temperature range (K), min. | 370.00 |
| Temperature range (K), max. | 502.15 |

## Sources

|   |   |
|---|---|
| <b>Joback Method:</b>   | <a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>   |
| <b>KDB:</b>   | <a href="https://www.therc.org/files/research/kdb/mol/mol872.mol">https://www.therc.org/files/research/kdb/mol/mol872.mol</a>   |
| <b>Crippen Method:</b>  | <a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>   |
| <b>KDB Vapor Pressure Data:</b>   | <a href="https://www.therc.org/research/kdb/hcprop/showprop.php?cmpid=872">https://www.therc.org/research/kdb/hcprop/showprop.php?cmpid=872</a>   |
| <b>Effect of Isomers on Partition Coefficients for Phenolic Compounds</b>             | <a href="https://www.doi.org/10.1021/je100016z">https://www.doi.org/10.1021/je100016z</a>   |
| <b>NIST Webbook - methylimidazolium Hexafluorophosphate + Water Two-Phase System:</b> | <a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C620177&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C620177&amp;Units=SI</a>   |
| <b>Liquid-Liquid Equilibrium Measurements for Model Systems</b>                       | <a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>   |
| <b>The Yaws Handbook of Vapor Pressure</b>  | <a href="https://www.doi.org/10.1021/acs.jced.6b00625">https://www.doi.org/10.1021/acs.jced.6b00625</a>   |
| <b>McGowan Method:</b>  | <a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a> |
| <b>Crippen Method:</b>  | <a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>   |

## Legend

|                 |   |
|-----------------|---|
| <b>chl:</b>     | Standard liquid enthalpy of combustion                    |
| <b>cpg:</b>     | Ideal gas heat capacity                                   |
| <b>dvisc:</b>   | Dynamic viscosity   |
| <b>gf:</b>      | Standard Gibbs free energy of formation                   |
| <b>hf:</b>      | Enthalpy of formation at standard conditions              |
| <b>hfl:</b>     | Liquid phase enthalpy of formation at standard conditions |
| <b>hfus:</b>    | Enthalpy of fusion at standard conditions                 |
| <b>hvap:</b>    | Enthalpy of vaporization at standard conditions           |
| <b>hvapt:</b>   | Enthalpy of vaporization at a given temperature           |
| <b>log10ws:</b> | Log10 of Water solubility in mol/l                        |
| <b>logp:</b>    | Octanol/Water partition coefficient                       |
| <b>mcvol:</b>   | McGowan's characteristic volume                           |
| <b>pc:</b>      | Critical Pressure   |
| <b>pvap:</b>    | Vapor pressure  |
| <b>rho:</b>     | Liquid Density  |
| <b>rinpol:</b>  | Non-polar retention indices                               |
| <b>ripol:</b>   | Polar retention indices                                   |
| <b>tb:</b>      | Normal Boiling Point Temperature                          |
| <b>tbrp:</b>    | Boiling point at reduced pressure                         |
| <b>tc:</b>      | Critical Temperature                                      |

**tf:** Normal melting (fusion) point

**vc:** Critical Volume

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