

# 2-Butanone, 4-(diethylamino)-

<b>Other names:</b>	1-(Diethylamino)-3-butanone 4-(Diethylamino)-2-butanone 4-(N,N-Diethylamino)butan-2-one 4-(diethylamino)butan-2-one
<b>Inchi:</b>	InChI=1S/C8H17NO/c1-4-9(5-2)7-6-8(3)10/h4-7H2,1-3H3
<b>InchiKey:</b>	XLEOCTUCGZANAC-UHFFFAOYSA-N
<b>Formula:</b>	C8H17NO
<b>SMILES:</b>	CCN(CC)CCC(C)=O
<b>Mol. weight [g/mol]:</b>	143.23
<b>CAS:</b>	3299-38-5

## Physical Properties

Property code	Value	Unit	Source
gf	-1.66	kJ/mol	Joback Method
hf	-253.50	kJ/mol	Joback Method
hfus	21.10	kJ/mol	Joback Method
hvap	42.19	kJ/mol	Joback Method
log10ws	-1.02		Crippen Method
logp	1.307		Crippen Method
mcvol	135.130	ml/mol	McGowan Method
pc	2698.60	kPa	Joback Method
rinpol	1025.00		NIST Webbook
rinpol	1025.00		NIST Webbook
rinpol	1025.00		NIST Webbook
tb	448.75	K	Joback Method
tc	622.49	K	Joback Method
tf	262.32	K	Joback Method
vc	0.507	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	286.50	J/molxK	448.75	Joback Method
cpg	299.93	J/molxK	477.71	Joback Method

cpg	312.79	J/mol×K	506.66	Joback Method
cpg	325.09	J/mol×K	535.62	Joback Method
cpg	336.84	J/mol×K	564.58	Joback Method
cpg	348.07	J/mol×K	593.54	Joback Method
cpg	358.78	J/mol×K	622.49	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	350.20	K	2.10	NIST Webbook

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C3299385&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C3299385&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point

**vc:** Critical Volume

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