

# p-Mentha-3-en-8-ol

<b>Inchi:</b>	InChI=1S/C11H20O/c1-9-4-6-10(7-5-9)11(2,3)8-12/h6,9,12H,4-5,7-8H2,1-3H3
<b>InchiKey:</b>	GRWJGTVTFCENHD-UHFFFAOYSA-N
<b>Formula:</b>	C11H20O
<b>SMILES:</b>	CC1CC=C(C(C)(C)CO)CC1
<b>Mol. weight [g/mol]:</b>	168.28

## Physical Properties

Property code	Value	Unit	Source
gf	-47.46	kJ/mol	Joback Method
hf	-330.72	kJ/mol	Joback Method
hfus	13.59	kJ/mol	Joback Method
hvap	56.85	kJ/mol	Joback Method
log10ws	-2.96		Crippen Method
logp	2.751		Crippen Method
mcvol	156.560	ml/mol	McGowan Method
pc	2665.27	kPa	Joback Method
rinqol	1149.00		NIST Webbook
tb	563.72	K	Joback Method
tc	761.02	K	Joback Method
tf	297.63	K	Joback Method
vc	0.579	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	402.09	J/molxK	563.72	Joback Method
cpg	418.69	J/molxK	596.60	Joback Method
cpg	434.34	J/molxK	629.49	Joback Method
cpg	449.08	J/molxK	662.37	Joback Method
cpg	462.96	J/molxK	695.25	Joback Method
cpg	476.02	J/molxK	728.14	Joback Method
cpg	488.29	J/molxK	761.02	Joback Method
dvisc	0.0168577	Paxs	297.63	Joback Method
dvisc	0.0040314	Paxs	341.98	Joback Method

dvisc	0.0013389	Paxs	386.33	Joback Method
dvisc	0.0005580	Paxs	430.68	Joback Method
dvisc	0.0002739	Paxs	475.02	Joback Method
dvisc	0.0001518	Paxs	519.37	Joback Method
dvisc	0.0000923	Paxs	563.72	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R286463&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R286463&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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