

# Fumaric acid, 3,4-dimethoxyphenyl isohexyl ester

Inchi:	InChI=1S/C18H24O6/c1-13(2)6-5-11-23-17(19)9-10-18(20)24-14-7-8-15(21-3)16(12-14)2
InchiKey:	WFFUTSPRGPXKH-MDZDMXLPSA-N
Formula:	C18H24O6
SMILES:	COc1ccc(OC(=O)C=CC(=O)OCCCC(C)C)cc1OC
Mol. weight [g/mol]:	336.38

## Physical Properties

Property code	Value	Unit	Source
gf	-406.23	kJ/mol	Joback Method
hf	-843.36	kJ/mol	Joback Method
hfus	40.27	kJ/mol	Joback Method
hvap	81.96	kJ/mol	Joback Method
log10ws	-3.85		Crippen Method
logp	3.145		Crippen Method
mcvol	263.040	ml/mol	McGowan Method
pc	1565.99	kPa	Joback Method
rinqol	2536.00		NIST Webbook
tb	849.02	K	Joback Method
tc	1057.00	K	Joback Method
tf	512.78	K	Joback Method
vc	0.994	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	791.61	J/molxK	849.02	Joback Method
cpg	805.92	J/molxK	883.68	Joback Method
cpg	819.06	J/molxK	918.35	Joback Method
cpg	831.02	J/molxK	953.01	Joback Method
cpg	841.80	J/molxK	987.67	Joback Method
cpg	851.40	J/molxK	1022.34	Joback Method
cpg	859.81	J/molxK	1057.00	Joback Method
dvisc	0.0003185	Paxs	512.78	Joback Method
dvisc	0.0001814	Paxs	568.82	Joback Method

dvisc	0.0001143	Paxs	624.86	Joback Method
dvisc	0.0000777	Paxs	680.90	Joback Method
dvisc	0.0000560	Paxs	736.94	Joback Method
dvisc	0.0000423	Paxs	792.98	Joback Method
dvisc	0.0000331	Paxs	849.02	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U348168&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U348168&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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