

Fumaric acid, di(2,4,6-trichlorophenyl) ester

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|-----------------------------|--|
| Inchi: | InChI=1S/C16H6Cl6O4/c17-7-3-9(19)15(10(20)4-7)25-13(23)1-2-14(24)26-16-11(21)5-8 |
| InchiKey: | PVAYWDBKLFNXLA-OWOJBTEDSA-N |
| Formula: | C16H6Cl6O4 |
| SMILES: | O=C(C=CC(=O)Oc1c(Cl)cc(Cl)cc1Cl)Oc1c(Cl)cc(Cl)cc1Cl |
| Mol. weight [g/mol]: | 474.93 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|----------------------|----------------|
| gf | -208.32 | kJ/mol | Joback Method |
| hf | -436.15 | kJ/mol | Joback Method |
| hfus | 53.90 | kJ/mol | Joback Method |
| hvap | 104.31 | kJ/mol | Joback Method |
| log10ws | -7.71 | | Crippen Method |
| logp | 6.674 | | Crippen Method |
| mcvol | 272.800 | ml/mol | McGowan Method |
| pc | 1957.87 | kPa | Joback Method |
| rinsol | 3081.00 | | NIST Webbook |
| tb | 1030.04 | K | Joback Method |
| tc | 1290.67 | K | Joback Method |
| tf | 716.80 | K | Joback Method |
| vc | 1.038 | m ³ /kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|-----------|---------|-----------------|---------------|
| cpg | 627.63 | J/molxK | 1030.04 | Joback Method |
| cpg | 632.60 | J/molxK | 1073.48 | Joback Method |
| cpg | 636.51 | J/molxK | 1116.92 | Joback Method |
| cpg | 639.40 | J/molxK | 1160.36 | Joback Method |
| cpg | 641.30 | J/molxK | 1203.79 | Joback Method |
| cpg | 642.23 | J/molxK | 1247.23 | Joback Method |
| cpg | 642.22 | J/molxK | 1290.67 | Joback Method |
| dvisc | 0.0001564 | Paxs | 716.80 | Joback Method |
| dvisc | 0.0001142 | Paxs | 769.01 | Joback Method |

| | | | | |
|-------|-----------|------|---------|---------------|
| dvisc | 0.0000868 | Paxs | 821.21 | Joback Method |
| dvisc | 0.0000682 | Paxs | 873.42 | Joback Method |
| dvisc | 0.0000550 | Paxs | 925.63 | Joback Method |
| dvisc | 0.0000455 | Paxs | 977.83 | Joback Method |
| dvisc | 0.0000383 | Paxs | 1030.04 | Joback Method |

Sources

| | |
|------------------------|---|
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci9903071 |
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=U348280&Units=SI |

Legend

| | |
|-----------------|---|
| cpg: | Ideal gas heat capacity |
| dvisc: | Dynamic viscosity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| rinpol: | Non-polar retention indices |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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