## Heptane, 1-bromo-

Other names: 1-Bromoheptane

Heptyl bromide

N-HEPTYL BROMIDE

**Inchi:** InChl=1S/C7H15Br/c1-2-3-4-5-6-7-8/h2-7H2,1H3

InchiKey: LSXKDWGTSHCFPP-UHFFFAOYSA-N

Formula: C7H15Br SMILES: CCCCCCBr

**Mol. weight [g/mol]:** 179.10 **CAS:** 629-04-9

### **Physical Properties**

Property code	Value	Unit	Source
chl	-4679.90 ± 1.60	kJ/mol	NIST Webbook
gf	22.38	kJ/mol	Joback Method
hf	-170.20 ± 2.20	kJ/mol	NIST Webbook
hfl	-218.60 ± 1.60	kJ/mol	NIST Webbook
hfus	19.17	kJ/mol	Joback Method
hvap	50.20	kJ/mol	NIST Webbook
hvap	50.80 ± 0.10	kJ/mol	NIST Webbook
hvap	50.61	kJ/mol	NIST Webbook
hvap	50.40 ± 0.20	kJ/mol	NIST Webbook
hvap	50.79 ± 0.10	kJ/mol	NIST Webbook
hvap	50.40 ± 0.20	kJ/mol	NIST Webbook
ie	9.81	eV	NIST Webbook
log10ws	-4.43		Estimated Solubility Method
log10ws	-4.43		Aqueous Solubility Prediction Method
logp	3.352		Crippen Method
mcvol	126.990	ml/mol	McGowan Method
рс	3015.64	kPa	Joback Method
rinpol	1061.00		NIST Webbook
rinpol	1027.00		NIST Webbook
rinpol	1027.00		NIST Webbook
rinpol	1030.00		NIST Webbook
rinpol	1040.00		NIST Webbook
rinpol	1031.00		NIST Webbook
rinpol	1067.00		NIST Webbook

rinpol	1036.00		NIST Webbook
rinpol	1018.00		NIST Webbook
rinpol	1042.00		NIST Webbook
rinpol	1036.00		NIST Webbook
ripol	1282.00		NIST Webbook
ripol	1266.00		NIST Webbook
ripol	1262.00		NIST Webbook
ripol	1268.00		NIST Webbook
ripol	1256.00		NIST Webbook
ripol	1256.00		NIST Webbook
ripol	1252.00		NIST Webbook
ripol	1262.00		NIST Webbook
tb	451.70	K	NIST Webbook
tb	453.00	K	NIST Webbook
tb	453.20	K	NIST Webbook
tb	453.15	K	KDB
tc	608.13	K	Joback Method
tf	214.40 ± 0.30	K	NIST Webbook
tf	217.10 ± 0.25	K	NIST Webbook
tf	215.82	К	Aqueous Solubility Prediction Method
tf	214.90 ± 0.40	K	NIST Webbook
VC	0.489	m3/kmol	Joback Method
		<u> </u>	

# **Temperature Dependent Properties**

Property code	Value	Unit	Temperature [K]	Source
cpg	236.26	J/mol×K	425.72	Joback Method
cpg	248.15	J/mol×K	456.12	Joback Method
cpg	259.52	J/mol×K	486.52	Joback Method
cpg	270.39	J/mol×K	516.93	Joback Method
cpg	280.77	J/mol×K	547.33	Joback Method
cpg	290.67	J/mol×K	577.73	Joback Method
cpg	300.13	J/mol×K	608.13	Joback Method
cpl	247.90	J/mol×K	298.15	NIST Webbook
cpl	251.24	J/mol×K	303.15	Isobaric heat capacity, isothermal compressibility and fluctuational properties of 1-bromoalkanes

cpl	253.07	J/mol×K	308.15	Isobaric heat capacity, isothermal compressibility and fluctuational properties of 1-bromoalkanes	
cpl	254.91	J/mol×K	313.15	Isobaric heat capacity, isothermal compressibility and fluctuational properties of 1- bromoalkanes	
cpl	249.39	J/mol×K	298.15	Isobaric heat capacity, isothermal compressibility and fluctuational properties of 1-bromoalkanes	
cpl	258.58	J/mol×K	323.15	Isobaric heat capacity, isothermal compressibility and fluctuational properties of 1- bromoalkanes	
cpl	260.43	J/mol×K	328.15	Isobaric heat capacity, isothermal compressibility and fluctuational properties of 1-bromoalkanes	
cpl	262.27	J/mol×K	333.15	Isobaric heat capacity, isothermal compressibility and fluctuational properties of 1- bromoalkanes	
cpl	264.10	J/mol×K	338.15	Isobaric heat capacity, isothermal compressibility and fluctuational properties of 1-bromoalkanes	
cpl	265.94	J/mol×K	343.15	Isobaric heat capacity, isothermal compressibility and fluctuational properties of 1- bromoalkanes	
cpl	267.77	J/mol×K	348.15	Isobaric heat capacity, isothermal compressibility and fluctuational properties of 1-bromoalkanes	

cpl	269.61	J/mol×K	353.15	Isobaric heat capacity, isothermal compressibility and fluctuational properties of 1- bromoalkanes	
cpl	271.46	J/mol×K	358.15	Isobaric heat capacity, isothermal compressibility and fluctuational properties of 1-bromoalkanes	
cpl	273.29	J/mol×K	363.15	Isobaric heat capacity, isothermal compressibility and fluctuational properties of 1- bromoalkanes	
cpl	275.13	J/mol×K	368.15	Isobaric heat capacity, isothermal compressibility and fluctuational properties of 1-bromoalkanes	
cpl	276.98	J/mol×K	373.15	Isobaric heat capacity, isothermal compressibility and fluctuational properties of 1-bromoalkanes	
cpl	278.80	J/mol×K	378.15	Isobaric heat capacity, isothermal compressibility and fluctuational properties of 1-bromoalkanes	
cpl	280.65	J/mol×K	383.15	Isobaric heat capacity, isothermal compressibility and fluctuational properties of 1-bromoalkanes	
срІ	282.47	J/mol×K	388.15	Isobaric heat capacity, isothermal compressibility and fluctuational properties of 1-bromoalkanes	

cpl	284.32	J/mol×K	393.15	Isobaric heat capacity, isothermal compressibility and fluctuational properties of 1- bromoalkanes	
cpl	286.16	J/mol×K	398.15	Isobaric heat capacity, isothermal compressibility and fluctuational properties of 1-bromoalkanes	
cpl	287.99	J/mol×K	403.15	Isobaric heat capacity, isothermal compressibility and fluctuational properties of 1- bromoalkanes	
cpl	289.83	J/mol×K	408.15	Isobaric heat capacity, isothermal compressibility and fluctuational properties of 1- bromoalkanes	
cpl	291.68	J/mol×K	413.15	Isobaric heat capacity, isothermal compressibility and fluctuational properties of 1- bromoalkanes	
cpl	293.51	J/mol×K	418.15	Isobaric heat capacity, isothermal compressibility and fluctuational properties of 1- bromoalkanes	
cpl	295.35	J/mol×K	423.15	Isobaric heat capacity, isothermal compressibility and fluctuational properties of 1-bromoalkanes	
cpl	244.73	J/mol×K	284.15	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis	

cpl	245.56	J/mol×K	286.65	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis	
cpl	246.44	J/mol×K	289.15	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis	
cpl	247.28	J/mol×K	291.65	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis	
cpl	248.12	J/mol×K	294.15	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis	

cpl	248.93	J/mol×K	296.65	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis	
cpl	249.70	J/mol×K	299.15	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis	
cpl	250.38	J/mol×K	301.65	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis	
cpl	251.56	J/mol×K	304.15	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis	

cpl	252.33	J/mol×K	306.65	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis	
cpl	253.19	J/mol×K	309.15	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis	
cpl	254.14	J/mol×K	311.65	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis	
cpl	255.00	J/mol×K	314.15	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis	

cpl	255.97	J/mol×K	316.65	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis	
cpl	256.84	J/mol×K	319.15	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis	
cpl	257.85	J/mol×K	321.65	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis	
cpl	258.76	J/mol×K	324.15	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis	

cpl	259.71	J/mol×K	326.65	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis	
cpl	260.68	J/mol×K	329.15	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis	
cpl	261.61	J/mol×K	331.65	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis	
cpl	262.50	J/mol×K	334.15	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis	

cpl	262.97	J/mol×K	336.65	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis	
cpl	264.64	J/mol×K	339.15	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis	
cpl	265.49	J/mol×K	341.65	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis	
cpl	266.52	J/mol×K	344.15	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis	

срІ	267.46	J/mol×K	346.65	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis	
cpl	268.47	J/mol×K	349.15	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis	
cpl	269.51	J/mol×K	351.65	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis	
cpl	270.03	J/mol×K	353.15	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis	
cpl	256.75	J/mol×K	318.15	Isobaric heat capacity, isothermal compressibility and fluctuational properties of 1-bromoalkanes	
dvisc	0.0041928	Paxs	228.45	Joback Method	

dvisc	0.0021013	Paxs	261.33	Joback Method	
dvisc	0.0012290	Paxs	294.21	Joback Method	
dvisc	0.0008006	Paxs	327.09	Joback Method	
dvisc	0.0005640	Paxs	359.96	Joback Method	
dvisc	0.0004213	Paxs	392.84	Joback Method	
dvisc	0.0003293	Paxs	425.72	Joback Method	
hfust	21.76	kJ/mol	214.40	NIST Webbook	
hfust	21.76	kJ/mol	214.40	NIST Webbook	
hvapt	47.00	kJ/mol	411.00	NIST Webbook	
hvapt	47.50	kJ/mol	408.00	NIST Webbook	
rhol	1084.50	kg/m3	343.15	Density of Some 1-Bromoalkanes within the Temperature Range from (243.15 to 423.15) K	
rhol	1169.60	kg/m3	263.15	Density of Some 1-Bromoalkanes within the Temperature Range from (243.15 to 423.15) K	
rhol	1191.00	kg/m3	243.15	Density of Some 1-Bromoalkanes within the Temperature Range from (243.15 to 423.15) K	
rhol	1026.60	kg/m3	393.15	Density of Some 1-Bromoalkanes within the Temperature Range from (243.15 to 423.15) K	
rhol	1132.90	kg/m3	298.15	Density of Some 1-Bromoalkanes within the Temperature Range from (243.15 to 423.15) K	
rhol	1106.00	kg/m3	323.15	Density of Some 1-Bromoalkanes within the Temperature Range from (243.15 to 423.15) K	
rhol	1159.20	kg/m3	273.15	Density of Some 1-Bromoalkanes within the Temperature Range from (243.15 to 423.15) K	

rhol	1148.70	kg/m3	283.15	Density of Some 1-Bromoalkanes within the Temperature Range from (243.15 to 423.15) K	
speedsl	748.09	m/s	423.22	Speed of Sound, Densities, and Isentropic Compressibilities of Liquid 1-Bromoalkanes at Temperatures from (243.15 to 423.15) K	
speedsl	1207.17	m/s	273.33	Speed of Sound, Densities, and Isentropic Compressibilities of Liquid 1-Bromoalkanes at Temperatures from (243.15 to 423.15) K	
speedsl	1108.66	m/s	303.19	Speed of Sound, Densities, and Isentropic Compressibilities of Liquid 1-Bromoalkanes at Temperatures from (243.15 to 423.15) K	
speedsl	1045.07	m/s	323.14	Speed of Sound, Densities, and Isentropic Compressibilities of Liquid 1-Bromoalkanes at Temperatures from (243.15 to 423.15) K	
speedsl	952.86	m/s	353.14	Speed of Sound, Densities, and Isentropic Compressibilities of Liquid 1-Bromoalkanes at Temperatures from (243.15 to 423.15) K	
speedsl	863.47	m/s	383.17	Speed of Sound, Densities, and Isentropic Compressibilities of Liquid 1-Bromoalkanes at Temperatures from (243.15 to 423.15) K	

speedsl	1308.35	m/s	243.74	Speed of Sound, Densities, and Isentropic Compressibilities of Liquid 1-Bromoalkanes at Temperatures from (243.15 to
				423.15) K

#### **Correlations**

Information	Value
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Property code	pvap
Equation	ln(Pvp) = A + B/(T + C)
Coeff. A	1.52868e+01
Coeff. B	-4.12208e+03
Coeff. C	-6.68200e+01
Temperature range (K), min.	341.64
Temperature range (K), max.	480.05

Information Value

Property code	pvap
Equation	$ln(Pvp) = A + B/T + C*ln(T) + D*T^2$
Coeff. A	8.18101e+01
Coeff. B	-8.70223e+03
Coeff. C	-9.62699e+00
Coeff. D	4.55484e-06
Temperature range (K), min.	217.05
Temperature range (K), max.	651.00

#### **Sources**

KDB:

Speed of Sound, Densities, and Isentropic Compressibilities of Liquid Pensity of Range at Transportances from withing he Temperature Range from (2018) Struck Frediction Method:

https://www.doi.org/10.1021/je900227j

https://www.doi.org/10.1021/je700015t

Isobaric heat capacity, isothermal compressibility and fluctuational properties by the books bromoalkanes:

mapo.,/ www.aon.org/ 10.1007/010700 010 2001 y

http://webbook.nist.gov/cgi/cbook.cgi?ID=C629049&Units=SI https://www.cheric.org/files/research/kdb/mol/mol1638.mol

**Estimated Solubility Method:** 

http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl\_file/ci034243xsi20040112\_053635.txt

The Yaws Handbook of Vapor https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure

Pressure: Crippen Method:

http://pubs.acs.org/doi/abs/10.1021/ci990307l

McGowan Method: http://link.springer.com/article/10.1007/BF02311772

Heat Capacities of 1-chloroalkanes and https://www.doi.org/10.1021/je049652j 1-bromoalkanes within the temperature

datas in Unit 204:15 K to 353.15 K. A group additivity and molecular commercially analysis.

https://en.wikipedia.org/wiki/Joback\_method

https://www.cheric.org/research/kdb/hcprop/showprop.php?cmpid=1638

### Legend

Standard liquid enthalpy of combustion chl:

Ideal gas heat capacity cpg:

Liquid phase heat capacity cpl:

dvisc: Dynamic viscosity

Standard Gibbs free energy of formation gf: hf: Enthalpy of formation at standard conditions

hfl: Liquid phase enthalpy of formation at standard conditions

hfus: Enthalpy of fusion at standard conditions hfust: Enthalpy of fusion at a given temperature

hvap: Enthalpy of vaporization at standard conditions hvapt: Enthalpy of vaporization at a given temperature

ie: Ionization energy

log10ws: Log10 of Water solubility in mol/l Octanol/Water partition coefficient logp: mcvol: McGowan's characteristic volume

Critical Pressure pc: pvap: Vapor pressure rhol: Liquid Density

rinpol: Non-polar retention indices

ripol: Polar retention indices speedsl: Speed of sound in fluid

tb: Normal Boiling Point Temperature

Critical Temperature tc:

tf: Normal melting (fusion) point

Critical Volume vc:

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