

Sulfur chloride (S2Cl2)

Other names:	Disulfur dichloride Sulfur chloride (S2Cl2) Chlorosulfane Sulfur chloride Sulfur subchloride Thiosulfurous dichloride S2Cl2 Chloride of sulfur Siarki chlorek Sulfur chloride(di) CISSCI disulphur dichloride
Inchi:	InChI=1S/Cl2S2/c1-3-4-2
InchiKey:	PXJJSXABGXMUSU-UHFFFAOYSA-N
Formula:	Cl2S2
SMILES:	CISSCI
Mol. weight [g/mol]:	135.04

Physical Properties

Property code	Value	Unit	Source
hfus	12.41	kJ/mol	Joback Method
ie	9.66 ± 0.03	eV	NIST Webbook
ie	11.30 ± 0.20	eV	NIST Webbook
ie	9.40	eV	NIST Webbook
logp	2.675		Crippen Method
mcvol	68.040	ml/mol	McGowan Method
tf	192.10 ± 0.20	K	NIST Webbook
tf	192.25 ± 0.10	K	NIST Webbook
tf	464.00 ± 3.00	K	NIST Webbook

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	77.79	J/mol×K	411.82	Joback Method

cpg	79.18	J/mol×K	452.96	Joback Method
cpg	80.56	J/mol×K	494.10	Joback Method
cpg	81.92	J/mol×K	535.25	Joback Method
cpg	83.23	J/mol×K	576.39	Joback Method
cpg	84.47	J/mol×K	617.53	Joback Method
cpg	85.62	J/mol×K	658.67	Joback Method
hvapt	41.10	kJ/mol	372.50	NIST Webbook

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Cheméo Relay (1.0):	
Cheméo Relay (1.0, beta):	
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C10025679&Units=SI
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
hfus:	Enthalpy of fusion at standard conditions
hvapt:	Enthalpy of vaporization at a given temperature
ie:	Ionization energy
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
tf:	Normal melting (fusion) point

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