

4-nitro-o-toluidine

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| Other names: | 2-methyl-4-nitroaniline 5-Nitro-2-aminotoluene aniline, 2-methyl-4-nitro- |
| Inchi: | InChI=1S/C7H8N2O2/c1-5-4-6(9(10)11)2-3-7(5)8/h2-4H,8H2,1H3 |
| InchiKey: | XTTIQGSLJBWVIV-UHFFFAOYSA-N |
| Formula: | C7H8N2O2 |
| SMILES: | Cc1cc([N+](=O)[O-])ccc1N |
| Mol. weight [g/mol]: | 152.15 |
| CAS: | 99-52-5 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|---------|----------------|
| gf | 203.21 | kJ/mol | Joback Method |
| hf | 48.81 | kJ/mol | Joback Method |
| hfus | 23.71 | kJ/mol | Joback Method |
| hvap | 62.01 | kJ/mol | Joback Method |
| log10ws | -2.23 | | Crippen Method |
| logp | 1.485 | | Crippen Method |
| mcvol | 113.130 | ml/mol | McGowan Method |
| pc | 4328.25 | kPa | Joback Method |
| tb | 620.57 | K | Joback Method |
| tc | 879.01 | K | Joback Method |
| tf | 446.98 | K | Joback Method |
| vc | 0.430 | m3/kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|--------|---------|-----------------|---------------|
| cpg | 314.89 | J/mol×K | 835.94 | Joback Method |
| cpg | 269.53 | J/mol×K | 620.57 | Joback Method |
| cpg | 280.15 | J/mol×K | 663.64 | Joback Method |
| cpg | 289.96 | J/mol×K | 706.72 | Joback Method |
| cpg | 299.00 | J/mol×K | 749.79 | Joback Method |
| cpg | 307.29 | J/mol×K | 792.87 | Joback Method |

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|-----|--------|---------|--------|---------------|
| cpg | 321.82 | J/mol×K | 879.01 | Joback Method |
| cps | 217.10 | J/mol×K | 323.00 | NIST Webbook |

Sources

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|--|---|
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci990307i |
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |
| Solubility and solution thermodynamics of 2-nitro-o-toluidine in mixed solvents at elevated temperatures. (Methyl acetate + (methanol, ethanol, n-propanol and isopropanol): McGowan Method: | https://www.doi.org/10.1016/j.jct.2016.10.037 |
| Solubility and preferential solvation of 2-nitro-o-toluidine in mixed solvents at elevated temperatures. (Methyl acetate + (methanol, ethanol, n-propanol and isopropanol): McGowan Method: | https://www.doi.org/10.1016/j.jct.2017.05.044 |
| NIST Webbook: | https://en.wikipedia.org/wiki/Joback_method |
| NIST Webbook: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=C99525&Units=SI |

Legend

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|----------|---|
| cpg: | Ideal gas heat capacity |
| cps: | Solid phase heat capacity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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