

# Cyclohexene, 4,4-dimethyl-

<b>Other names:</b>	1-Cyclohexene, 4,4-dimethyl- 4,4-DIMETHYL-1-CYCLOHEXENE 4,4-Dimethylcyclohexene
<b>Inchi:</b>	InChI=1S/C8H14/c1-8(2)6-4-3-5-7-8/h3-4H,5-7H2,1-2H3
<b>InchiKey:</b>	DRORSPJLYCDESA-UHFFFAOYSA-N
<b>Formula:</b>	C8H14
<b>SMILES:</b>	CC1(C)CC=CCC1
<b>Mol. weight [g/mol]:</b>	110.20
<b>CAS:</b>	14072-86-7

## Physical Properties

Property code	Value	Unit	Source
gf	65.40	kJ/mol	Joback Method
hf	-81.11	kJ/mol	Joback Method
hfus	3.24	kJ/mol	Joback Method
hvap	32.97	kJ/mol	Joback Method
log10ws	-2.68		Crippen Method
logp	2.753		Crippen Method
mcvol	108.420	ml/mol	McGowan Method
pc	3460.21	kPa	Joback Method
tb	394.00 ± 5.00	K	NIST Webbook
tc	613.90	K	Joback Method
tf	211.96	K	Joback Method
vc	0.401	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	200.60	J/mol×K	401.39	Joback Method
cpg	217.67	J/mol×K	436.81	Joback Method
cpg	233.51	J/mol×K	472.23	Joback Method
cpg	248.21	J/mol×K	507.65	Joback Method
cpg	261.88	J/mol×K	543.07	Joback Method
cpg	274.63	J/mol×K	578.48	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.35516e+01
Coeff. B	-3.01564e+03
Coeff. C	-5.64260e+01
Temperature range (K), min.	283.78
Temperature range (K), max.	422.40

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>KDB:</b>	<a href="https://www.thermo.com/files/research/kdb/mol/mol645.mol">https://www.thermo.com/files/research/kdb/mol/mol645.mol</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C14072867&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C14072867&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure

**tb:** Normal Boiling Point Temperature  
**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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