

# Naphthalene, 1,2,3,4-tetrahydro-1,8-dimethyl-

<b>Other names:</b>	1,8-Dimethyl-[1,2,3,4-tetrahydronaphthalene] 1,8-Dimethyltetralin
<b>Inchi:</b>	InChI=1S/C12H16/c1-9-5-3-7-11-8-4-6-10(2)12(9)11/h3,5,7,10H,4,6,8H2,1-2H3
<b>InchiKey:</b>	KVNZVXVZQFTXNA-UHFFFAOYSA-N
<b>Formula:</b>	C12H16
<b>SMILES:</b>	Cc1cccc2c1C(C)CCC2
<b>Mol. weight [g/mol]:</b>	160.26
<b>CAS:</b>	25419-33-4

## Physical Properties

Property code	Value	Unit	Source
gf	191.96	kJ/mol	Joback Method
hf	-10.78	kJ/mol	Joback Method
hfus	16.13	kJ/mol	Joback Method
hvap	45.99	kJ/mol	Joback Method
log10ws	-3.83		Crippen Method
logp	3.435		Crippen Method
mvol	145.320	ml/mol	McGowan Method
pc	2729.71	kPa	Joback Method
rinpol	1318.00		NIST Webbook
tb	521.61	K	Joback Method
tc	747.23	K	Joback Method
tf	290.88	K	Joback Method
vc	0.548	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	333.07	J/mol×K	521.61	Joback Method
cpg	351.33	J/mol×K	559.21	Joback Method
cpg	368.45	J/mol×K	596.82	Joback Method
cpg	384.49	J/mol×K	634.42	Joback Method
cpg	399.49	J/mol×K	672.02	Joback Method
cpg	413.53	J/mol×K	709.63	Joback Method

cpg	426.65	J/mol×K	747.23	Joback Method
dvisc	0.0016585	Paxs	290.88	Joback Method
dvisc	0.0010799	Paxs	329.33	Joback Method
dvisc	0.0007692	Paxs	367.79	Joback Method
dvisc	0.0005842	Paxs	406.25	Joback Method
dvisc	0.0004653	Paxs	444.70	Joback Method
dvisc	0.0003843	Paxs	483.16	Joback Method
dvisc	0.0003265	Paxs	521.61	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C25419334&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C25419334&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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