

# Cyclopentane, 1-methyl-2-(4-methylpentyl)-, trans-

<b>Other names:</b>	trans-1-Methyl-2-(4-methylpentyl)cyclopentane
<b>Inchi:</b>	InChI=1S/C12H24/c1-10(2)6-4-8-12-9-5-7-11(12)3/h10-12H,4-9H2,1-3H3/t11-,12-/m1/s1
<b>InchiKey:</b>	YWCSZNJUSYMHHA-VXGBXAGGSA-N
<b>Formula:</b>	C12H24
<b>SMILES:</b>	CC(C)CCCC1CCCC1C
<b>Mol. weight [g/mol]:</b>	168.32
<b>CAS:</b>	66553-50-2

## Physical Properties

Property code	Value	Unit	Source
gf	76.56	kJ/mol	Joback Method
hf	-256.15	kJ/mol	Joback Method
hfus	18.32	kJ/mol	Joback Method
hvap	41.87	kJ/mol	Joback Method
log10ws	-4.02		Crippen Method
logp	4.249		Crippen Method
mcvol	169.080	ml/mol	McGowan Method
pc	2038.23	kPa	Joback Method
rinpol	1151.80		NIST Webbook
tb	484.13	K	Joback Method
tc	673.31	K	Joback Method
tf	216.66	K	Joback Method
vc	0.641	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	394.42	J/molxK	484.13	Joback Method
cpg	415.16	J/molxK	515.66	Joback Method
cpg	434.94	J/molxK	547.19	Joback Method
cpg	453.78	J/molxK	578.72	Joback Method
cpg	471.71	J/molxK	610.25	Joback Method
cpg	488.76	J/molxK	641.78	Joback Method
cpg	504.95	J/molxK	673.31	Joback Method

dvisc	0.0051591	Paxs	216.66	Joback Method
dvisc	0.0020997	Paxs	261.24	Joback Method
dvisc	0.0011106	Paxs	305.82	Joback Method
dvisc	0.0006908	Paxs	350.39	Joback Method
dvisc	0.0004783	Paxs	394.97	Joback Method
dvisc	0.0003568	Paxs	439.55	Joback Method
dvisc	0.0002809	Paxs	484.13	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C66553502&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C66553502&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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