

# 2-Pentene, 2,4-dimethyl-

<b>Other names:</b>	(CH <sub>3</sub> ) <sub>2</sub> CHCH=C(CH <sub>3</sub> ) <sub>2</sub> 2,4-Dimethyl-2-pentene 2,4-dimethylpent-2-ene
<b>Inchi:</b>	InChI=1S/C7H14/c1-6(2)5-7(3)4/h5-6H,1-4H3
<b>InchiKey:</b>	VVCFYASOGFVJFN-UHFFFAOYSA-N
<b>Formula:</b>	C <sub>7</sub> H <sub>14</sub>
<b>SMILES:</b>	CC(C)=CC(C)C
<b>Mol. weight [g/mol]:</b>	98.19
<b>CAS:</b>	625-65-0

## Physical Properties

Property code	Value	Unit	Source
affp	812.00	kJ/mol	NIST Webbook
basg	783.10	kJ/mol	NIST Webbook
chl	-4632.27 ± 0.84	kJ/mol	NIST Webbook
gf	77.29	kJ/mol	Joback Method
hf	-85.66	kJ/mol	Joback Method
hfus	9.25	kJ/mol	Joback Method
hvap	34.30	kJ/mol	NIST Webbook
hvap	34.30	kJ/mol	NIST Webbook
hvap	34.50 ± 1.50	kJ/mol	NIST Webbook
log10ws	-2.36		Crippen Method
logp	2.609		Crippen Method
mcvol	105.190	ml/mol	McGowan Method
pc	3002.44	kPa	Joback Method
rinpol	640.00		NIST Webbook
rinpol	640.00		NIST Webbook
rinpol	641.00		NIST Webbook
rinpol	641.00		NIST Webbook
rinpol	640.70		NIST Webbook
rinpol	643.00		NIST Webbook
rinpol	641.00		NIST Webbook
rinpol	641.00		NIST Webbook
rinpol	641.00		NIST Webbook
rinpol	654.00		NIST Webbook
rinpol	641.00		NIST Webbook
rinpol	641.00		NIST Webbook

rinpol	655.50		NIST Webbook
rinpol	656.00		NIST Webbook
rinpol	641.00		NIST Webbook
rinpol	640.00		NIST Webbook
rinpol	641.00		NIST Webbook
rinpol	655.00		NIST Webbook
rinpol	650.00		NIST Webbook
rinpol	649.00		NIST Webbook
rinpol	650.00		NIST Webbook
rinpol	655.00		NIST Webbook
rinpol	641.00		NIST Webbook
rinpol	641.00		NIST Webbook
rinpol	655.50		NIST Webbook
rinpol	640.30		NIST Webbook
rinpol	640.70		NIST Webbook
rinpol	648.40		NIST Webbook
rinpol	649.10		NIST Webbook
rinpol	641.20		NIST Webbook
rinpol	654.90		NIST Webbook
rinpol	641.00		NIST Webbook
rinpol	654.00		NIST Webbook
tb	363.16	K	Joback Method
tc	543.24	K	Joback Method
tf	145.85 ± 0.05	K	NIST Webbook
tf	145.45 ± 0.01	K	NIST Webbook
tf	145.42 ± 0.02	K	NIST Webbook
tf	145.41 ± 0.03	K	NIST Webbook
tf	145.45 ± 0.02	K	NIST Webbook
vc	0.403	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	215.38	J/mol×K	453.20	Joback Method
cpg	226.07	J/mol×K	483.22	Joback Method
cpg	236.27	J/mol×K	513.23	Joback Method
cpg	180.18	J/mol×K	363.16	Joback Method
cpg	192.45	J/mol×K	393.17	Joback Method
cpg	204.18	J/mol×K	423.19	Joback Method
cpg	246.01	J/mol×K	543.24	Joback Method
hvapt	35.20 ± 1.50	kJ/mol	286.50	NIST Webbook

hvapt	34.50	kJ/mol	324.50	NIST Webbook
hvapt	34.20	kJ/mol	324.50	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.37580e+01
Coeff. B	-2.74151e+03
Coeff. C	-5.64900e+01
Temperature range (K), min.	260.01
Temperature range (K), max.	381.07

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/T + C \cdot \ln(T) + D \cdot T^2$
Coeff. A	1.01167e+02
Coeff. B	-7.43140e+03
Coeff. C	-1.31513e+01
Coeff. D	1.24302e-05
Temperature range (K), min.	260.15
Temperature range (K), max.	380.15

## Sources

<b>KDB Vapor Pressure Data:</b>	<a href="https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=242">https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=242</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemed.com/doc/models/crippen_log10ws">https://www.chemed.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>KDB:</b>	<a href="https://www.thermo.com/files/research/kdb/mol/mol242.mol">https://www.thermo.com/files/research/kdb/mol/mol242.mol</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C625650&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C625650&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>

# Legend

<b>affp:</b>	Proton affinity
<b>basg:</b>	Gas basicity
<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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