

# Ketoprofen

**Other names:**

(+)«alpha»-(3-benzoylphenyl)propionic acid ((+)-ketoprofen)  
(+)«alpha»-(3-benzoylphenyl)propionic acid ((+)-ketoprofen)  
(. +/-)-3-Benzoyl-«alpha»-methylbenzeneacetic acid  
(. +/-)-3-Benzoyl-«alpha»-methylbenzeneacetic acid  
(. +/-)-m-Benzoylhydratropic acid  
19583 RP  
2-(3-Benzoylphenyl)propionic acid  
2-(m-Benzoylphenyl)propionic acid  
3-Benzoyl-«alpha»-methylbenzeneacetic acid  
3-Benzoyl-«alpha»-methylbenzeneacetic acid  
3-Benzoylhydratropic acid  
Alrheumat  
Alrheumun  
Aneol  
Benzeneacetic acid, 3-benzoyl-«alpha»-methyl-  
Benzeneacetic acid, 3-benzoyl-«alpha»-methyl-  
Capisten  
Dexal  
Epatec  
Fastum  
Iso-K  
Kefenid  
Ketoprofene  
Ketopron  
Lertus  
Menamin  
Meprofen  
Orudis  
Orugesic  
Oruvail  
Oscorel  
Profenid  
Propionic acid, 2-(3-benzoylphenyl)-  
RP-19583  
RU 4733  
Racemic ketoprofen  
Toprec  
Toprek  
m-Benzoylhydratropic acid

**Inchi:**

InChI=1S/C16H14O3/c1-11(16(18)19)13-8-5-9-14(10-13)15(17)12-6-3-2-4-7-12/h2-11H,1

**InchiKey:** DKYWVDODHFEZIM-UHFFFAOYSA-N  
**Formula:** C<sub>16</sub>H<sub>14</sub>O<sub>3</sub>  
**SMILES:** CC(C(=O)O)c1cccc(C(=O)c2ccccc2)c1  
**Mol. weight [g/mol]:** 254.28  
**CAS:** 22071-15-4

## Physical Properties

Property code	Value	Unit	Source
gf	-98.07	kJ/mol	Joback Method
hf	-294.65	kJ/mol	Joback Method
hfus	28.65	kJ/mol	Joback Method
hvap	86.21	kJ/mol	Joback Method
log10ws	-3.45		Aqueous Solubility Prediction Method
log10ws	-3.21		Aqueous Solubility Prediction Method
logp	3.106		Crippen Method
mcvol	197.790	ml/mol	McGowan Method
pc	2826.33	kPa	Joback Method
rinpola	2185.00		NIST Webbook
tb	823.30	K	Joback Method
tc	1052.13	K	Joback Method
tf	367.12	K	Aqueous Solubility Prediction Method
tf	367.01	K	Aqueous Solubility Prediction Method
vc	0.741	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	557.11	J/mol×K	823.30	Joback Method
cpg	568.42	J/mol×K	861.44	Joback Method
cpg	578.78	J/mol×K	899.58	Joback Method
cpg	588.24	J/mol×K	937.72	Joback Method
cpg	596.87	J/mol×K	975.85	Joback Method
cpg	604.74	J/mol×K	1013.99	Joback Method
cpg	611.91	J/mol×K	1052.13	Joback Method

dvisc	0.0001707	Paxs	595.18	Joback Method
dvisc	0.0003532	Paxs	538.15	Joback Method
dvisc	0.0008679	Paxs	481.12	Joback Method
dvisc	0.0000937	Paxs	652.21	Joback Method
dvisc	0.0000567	Paxs	709.24	Joback Method
dvisc	0.0000369	Paxs	766.27	Joback Method
dvisc	0.0000255	Paxs	823.30	Joback Method
hfust	37.30	kJ/mol	368.00	NIST Webbook
hfust	25.04	kJ/mol	369.00	NIST Webbook
hfust	28.23	kJ/mol	367.40	NIST Webbook
hsubt	110.10 ± 0.50	kJ/mol	353.00	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>Thermodynamic analysis of the solubility of ketoprofen in some binary hydro-organic solvent systems</b>	<a href="https://www.doi.org/10.1016/j.fluid.2010.03.031">https://www.doi.org/10.1016/j.fluid.2010.03.031</a>
<b>Solubility of Ketoprofen in Ethanol + Water</b>	<a href="https://www.doi.org/10.1021/je9002866">https://www.doi.org/10.1021/je9002866</a>
<b>The Solubility of Benzocaine, Lidocaine, and Procaine in Liquid and Supercritical Carbon Dioxide</b>	<a href="https://www.doi.org/10.1021/je034163p">https://www.doi.org/10.1021/je034163p</a>
<b>Solubility and Association of Ketoprofen and Aspirin in Supercritical CO<sub>2</sub> by Aqueous Solubility Prediction Method:</b>	<a href="https://www.doi.org/10.1021/acs.jced.5b00812">https://www.doi.org/10.1021/acs.jced.5b00812</a>
<b>Aqueous Solubility Prediction Method:</b>	<a href="http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDataset002.xlsx">http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDataset002.xlsx</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C22071154&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C22071154&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hsubt:</b>	Enthalpy of sublimation at a given temperature
<b>hvp:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mvol:</b>	McGowan's characteristic volume

<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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