

# Cyclobutane, 1,2-dichloro-1,2,3,3,4,4-hexafluoro-

<b>Other names:</b>	1,2-Dichloro-1,2,3,3,4,4-hexafluorocyclobutane 1,2-Dichlorohexafluorocyclobutane 1,2-Dichloroperfluorocyclobutane Cyclobutane, 1,2-dichlorohexafluoro-
<b>Inchi:</b>	InChI=1S/C4Cl2F6/c5-1(7)2(6,8)4(11,12)3(1,9)10
<b>InchiKey:</b>	LMHAGAHDHRQIMB-UHFFFAOYSA-N
<b>Formula:</b>	C4Cl2F6
<b>SMILES:</b>	FC1(F)C(F)(F)C(F)(Cl)C1(F)Cl
<b>Mol. weight [g/mol]:</b>	232.94
<b>CAS:</b>	356-18-3

## Physical Properties

Property code	Value	Unit	Source
gf	-1206.36	kJ/mol	Joback Method
hf	-1267.45	kJ/mol	Joback Method
hfus	7.05	kJ/mol	Joback Method
hvap	22.92	kJ/mol	Joback Method
log10ws	-3.25		Crippen Method
logp	3.080		Crippen Method
mcvol	91.460	ml/mol	McGowan Method
pc	3419.86	kPa	Joback Method
tb	332.70	K	NIST Webbook
tb	332.50 ± 0.50	K	NIST Webbook
tc	533.30	K	Joback Method
tf	295.52	K	Joback Method
vc	0.404	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	173.71	J/mol×K	359.36	Joback Method
cpg	184.75	J/mol×K	388.35	Joback Method
cpg	194.10	J/mol×K	417.34	Joback Method
cpg	201.97	J/mol×K	446.33	Joback Method

cpg	208.53	J/mol×K	475.32	Joback Method
cpg	213.96	J/mol×K	504.31	Joback Method
cpg	218.46	J/mol×K	533.30	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.52080e+01
Coeff. B	-3.16643e+03
Coeff. C	-3.34880e+01
Temperature range (K), min.	245.71
Temperature range (K), max.	353.44

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C356183&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C356183&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure

<b>pvap:</b>	Vapor pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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