

# (Z)-2,6-Dimethylocta-2,5,7-trien-4-one

<b>Other names:</b>	(Z)-Tagetenone cis-Tagetenone cis-Ocimenone (Z)-Ocimenone Ocimenone Z (Z)-Ocymenone (tagetenone) (Z)-Ocimenone (cis-ocimenone)
<b>Inchi:</b>	InChI=1S/C10H14O/c1-5-9(4)7-10(11)6-8(2)3/h5-7H,1H2,2-4H3/b9-7-
<b>InchiKey:</b>	XUINKEIPBTYUJP-CLFYSBASSA-N
<b>Formula:</b>	C10H14O
<b>SMILES:</b>	<chem>C=CC(C)=CC(=O)C=C(C)C</chem>
<b>Mol. weight [g/mol]:</b>	150.22
<b>CAS:</b>	33746-71-3

## Physical Properties

Property code	Value	Unit	Source
gf	135.58	kJ/mol	Joback Method
hf	-22.02	kJ/mol	Joback Method
hfus	19.76	kJ/mol	Joback Method
hvap	44.01	kJ/mol	Joback Method
log10ws	-2.85		Crippen Method
logp	2.654		Crippen Method
mcvol	140.430	ml/mol	McGowan Method
pc	2605.74	kPa	Joback Method
rinpol	1235.00		NIST Webbook
rinpol	1184.00		NIST Webbook
rinpol	1234.00		NIST Webbook
rinpol	1228.00		NIST Webbook
rinpol	1231.00		NIST Webbook
rinpol	1232.00		NIST Webbook
rinpol	1216.00		NIST Webbook
rinpol	1231.00		NIST Webbook
rinpol	1229.00		NIST Webbook
rinpol	1211.00		NIST Webbook
rinpol	1231.00		NIST Webbook
rinpol	1221.00		NIST Webbook
rinpol	1229.00		NIST Webbook

rinpol	1229.00		NIST Webbook
rinpol	1209.00		NIST Webbook
rinpol	1184.00		NIST Webbook
rinpol	1206.00		NIST Webbook
rinpol	1231.00		NIST Webbook
ripol	1704.00		NIST Webbook
ripol	1690.00		NIST Webbook
tb	486.83	K	Joback Method
tc	686.94	K	Joback Method
tf	212.55	K	Joback Method
vc	0.544	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	290.46	J/mol×K	486.83	Joback Method
cpg	304.04	J/mol×K	520.18	Joback Method
cpg	316.83	J/mol×K	553.53	Joback Method
cpg	328.85	J/mol×K	586.88	Joback Method
cpg	340.17	J/mol×K	620.23	Joback Method
cpg	350.82	J/mol×K	653.59	Joback Method
cpg	360.87	J/mol×K	686.94	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C33746713&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C33746713&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions

<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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