

# «beta»-lonyl acetate

|                      |                                                                                   |
|----------------------|-----------------------------------------------------------------------------------|
| Other names:         | 3-(2,6,6-trimethyl-1-cyclohexen-1-yl)propen-1-yl acetate                          |
| Inchi:               | InChI=1S/C15H24O2/c1-11-7-6-10-15(4,5)14(11)9-8-12(2)17-13(3)16/h8-9,12H,6-7,10H2 |
| InchiKey:            | WODKSVNXBYBTQC-CMDGGGOBGSA-N                                                      |
| Formula:             | C15H24O2                                                                          |
| SMILES:              | CC(=O)OC(C)C=CC1=C(C)CCCC1(C)C                                                    |
| Mol. weight [g/mol]: | 236.35                                                                            |
| CAS:                 | 22030-19-9                                                                        |

## Physical Properties

| Property code | Value   | Unit    | Source         |
|---------------|---------|---------|----------------|
| gf            | -51.06  | kJ/mol  | Joback Method  |
| hf            | -381.39 | kJ/mol  | Joback Method  |
| hfus          | 20.05   | kJ/mol  | Joback Method  |
| hvap          | 58.60   | kJ/mol  | Joback Method  |
| log10ws       | -4.44   |         | Crippen Method |
| logp          | 4.021   |         | Crippen Method |
| mcvol         | 210.190 | ml/mol  | McGowan Method |
| pc            | 1896.95 | kPa     | Joback Method  |
| rinpol        | 1525.00 |         | NIST Webbook   |
| rinpol        | 1525.00 |         | NIST Webbook   |
| rinpol        | 1525.00 |         | NIST Webbook   |
| tb            | 651.52  | K       | Joback Method  |
| tc            | 865.00  | K       | Joback Method  |
| tf            | 367.97  | K       | Joback Method  |
| vc            | 0.790   | m3/kmol | Joback Method  |

## Temperature Dependent Properties

| Property code | Value  | Unit    | Temperature [K] | Source        |
|---------------|--------|---------|-----------------|---------------|
| cpg           | 568.90 | J/mol×K | 651.52          | Joback Method |
| cpg           | 587.72 | J/mol×K | 687.10          | Joback Method |
| cpg           | 605.60 | J/mol×K | 722.68          | Joback Method |
| cpg           | 622.67 | J/mol×K | 758.26          | Joback Method |
| cpg           | 639.04 | J/mol×K | 793.84          | Joback Method |

|     |        |         |        |               |
|-----|--------|---------|--------|---------------|
| cpg | 654.81 | J/mol×K | 829.42 | Joback Method |
| cpg | 670.10 | J/mol×K | 865.00 | Joback Method |

## Sources

|                        |                                                                                                                                               |
|------------------------|-----------------------------------------------------------------------------------------------------------------------------------------------|
| <b>Crippen Method:</b> | <a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>                                     |
| <b>Crippen Method:</b> | <a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>                             |
| <b>Joback Method:</b>  | <a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>                                         |
| <b>McGowan Method:</b> | <a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>                         |
| <b>NIST Webbook:</b>   | <a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C22030199&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C22030199&amp;Units=SI</a> |

## Legend

|                 |                                                 |
|-----------------|-------------------------------------------------|
| <b>cpg:</b>     | Ideal gas heat capacity                         |
| <b>gf:</b>      | Standard Gibbs free energy of formation         |
| <b>hf:</b>      | Enthalpy of formation at standard conditions    |
| <b>hfus:</b>    | Enthalpy of fusion at standard conditions       |
| <b>hvac:</b>    | Enthalpy of vaporization at standard conditions |
| <b>log10ws:</b> | Log10 of Water solubility in mol/l              |
| <b>logp:</b>    | Octanol/Water partition coefficient             |
| <b>mcvol:</b>   | McGowan's characteristic volume                 |
| <b>pc:</b>      | Critical Pressure                               |
| <b>rinpol:</b>  | Non-polar retention indices                     |
| <b>tb:</b>      | Normal Boiling Point Temperature                |
| <b>tc:</b>      | Critical Temperature                            |
| <b>tf:</b>      | Normal melting (fusion) point                   |
| <b>vc:</b>      | Critical Volume                                 |

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