

1-Butene, 2-chloro-3-methyl-

Other names:	2-Chloro-3-methyl-1-butene CH ₂ =C(Cl)CH(CH ₃) ₂
Inchi:	InChI=1S/C5H9Cl/c1-4(2)5(3)6/h4H,3H2,1-2H3
InchiKey:	RBSYGFLXVMWYGD-UHFFFAOYSA-N
Formula:	C ₅ H ₉ Cl
SMILES:	C=C(Cl)C(C)C
Mol. weight [g/mol]:	104.58
CAS:	17773-64-7

Physical Properties

Property code	Value	Unit	Source
gf	56.14	kJ/mol	Joback Method
hf	-51.91	kJ/mol	Joback Method
hfus	6.79	kJ/mol	Joback Method
hvap	30.13	kJ/mol	Joback Method
log10ws	-2.18		Crippen Method
logp	2.395		Crippen Method
mcvol	89.250	ml/mol	McGowan Method
pc	3543.08	kPa	Joback Method
rinpol	626.00		NIST Webbook
rinpol	629.00		NIST Webbook
rinpol	626.00		NIST Webbook
tb	347.35	K	Joback Method
tc	531.71	K	Joback Method
tf	145.31	K	Joback Method
vc	0.341	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	137.98	J/mol×K	347.35	Joback Method
cpg	146.76	J/mol×K	378.08	Joback Method
cpg	155.15	J/mol×K	408.80	Joback Method
cpg	163.16	J/mol×K	439.53	Joback Method

cpg	170.79	J/mol×K	470.26	Joback Method
cpg	178.07	J/mol×K	500.99	Joback Method
cpg	185.00	J/mol×K	531.71	Joback Method

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.36588e+01
Coeff. B	-3.02744e+03
Coeff. C	-4.21060e+01
Temperature range (K), min.	268.52
Temperature range (K), max.	404.79

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C17773647&Units=SI
The Yaws Handbook of Vapor Pressure:	https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure

pvap:	Vapor pressure
rinpola:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/32-113-0/1-Butene-2-chloro-3-methyl.pdf>

Generated by Cheméo on 2024-04-19 17:47:10.00544035 +0000 UTC m=+15838078.926017665.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.