

# Butanoic acid, 2-chloro, (E)-3-hexenyl ester

<b>Inchi:</b>	InChI=1S/C10H17ClO2/c1-3-5-6-7-8-13-10(12)9(11)4-2/h5-6,9H,3-4,7-8H2,1-2H3/b6-5+
<b>InchiKey:</b>	WKCOAKPPMVZACK-AATRIKPKSA-N
<b>Formula:</b>	C10H17ClO2
<b>SMILES:</b>	CCC=CCCOC(=O)C(Cl)CC
<b>Mol. weight [g/mol]:</b>	204.69

## Physical Properties

Property code	Value	Unit	Source
gf	-134.75	kJ/mol	Joback Method
hf	-398.33	kJ/mol	Joback Method
hfus	25.32	kJ/mol	Joback Method
hvap	50.97	kJ/mol	Joback Method
log10ws	-2.99		Crippen Method
logp	2.903		Crippen Method
mcvol	167.140	ml/mol	McGowan Method
pc	2248.26	kPa	Joback Method
ripol	1294.00		NIST Webbook
ripol	1315.00		NIST Webbook
ripol	1305.00		NIST Webbook
ripol	1320.00		NIST Webbook
ripol	1750.00		NIST Webbook
ripol	1765.00		NIST Webbook
ripol	1700.00		NIST Webbook
ripol	1700.00		NIST Webbook
ripol	1797.00		NIST Webbook
tb	545.64	K	Joback Method
tc	734.12	K	Joback Method
tf	284.46	K	Joback Method
vc	0.642	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	380.20	J/mol×K	545.64	Joback Method

cpg	393.74	J/molxK	577.05	Joback Method
cpg	406.63	J/molxK	608.47	Joback Method
cpg	418.89	J/molxK	639.88	Joback Method
cpg	430.56	J/molxK	671.29	Joback Method
cpg	441.63	J/molxK	702.71	Joback Method
cpg	452.15	J/molxK	734.12	Joback Method
dvisc	0.0036477	Paxs	284.46	Joback Method
dvisc	0.0015709	Paxs	327.99	Joback Method
dvisc	0.0008241	Paxs	371.52	Joback Method
dvisc	0.0004950	Paxs	415.05	Joback Method
dvisc	0.0003275	Paxs	458.58	Joback Method
dvisc	0.0002328	Paxs	502.11	Joback Method
dvisc	0.0001748	Paxs	545.64	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R28675&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R28675&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

**tf:** Normal melting (fusion) point

**vc:** Critical Volume

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