

# Chloropropylate

<b>Other names:</b>	1-Methylethyl 4-chloro-«alpha»-(4-chlorophenyl)-«alpha»-hydroxybenzeneacetate 1-Methylethyl 4-chloro-«alpha»-(4-chlorophenyl)-«alpha»-hydroxybenzeneacetate Acaralate Benzeneacetic acid, 4-chloro-«alpha»-(4-chlorophenyl)-«alpha»-hydroxy-, 1-methylethyl ester Benzeneacetic acid, 4-chloro-«alpha»-(4-chlorophenyl)-«alpha»-hydroxy-, 1-methylethyl ester Benzilic acid, 4,4'-dichloro-, isopropyl ester Chlormite Chlorpropylat Chlorpropylate ENT 26,999 G 24,163 GEIGY 24163 Gesakur Isopropyl 4,4'-Dichlorobenzilate Isopropylester kyseliny 4,4'-dichlorbenzilove Rospan Rospin Rospine propan-2-yl 2,2-bis(4-chlorophenyl)-2-hydroxyacetate
<b>Inchi:</b>	InChI=1S/C17H16Cl2O3/c1-11(2)22-16(20)17(21,12-3-7-14(18)8-4-12)13-5-9-15(19)10-6
<b>InchiKey:</b>	AXGUBXVWZBFQGA-UHFFFAOYSA-N
<b>Formula:</b>	C17H16Cl2O3
<b>SMILES:</b>	CC(C)OC(=O)C(O)(c1ccc(Cl)cc1)c1ccc(Cl)cc1
<b>Mol. weight [g/mol]:</b>	339.21
<b>CAS:</b>	5836-10-2

## Physical Properties

Property code	Value	Unit	Source
gf	-96.38	kJ/mol	Joback Method
hf	-386.63	kJ/mol	Joback Method
hfus	31.42	kJ/mol	Joback Method
hvap	92.23	kJ/mol	Joback Method
log10ws	-4.53		Aqueous Solubility Prediction Method
log10ws	-4.53		Estimated Solubility Method
logp	4.181		Crippen Method

mvol	240.660	ml/mol	McGowan Method
pc	2210.37	kPa	Joback Method
rinpol	2189.00		NIST Webbook
rinpol	2189.00		NIST Webbook
ripol	3337.00		NIST Webbook
tb	891.34	K	Joback Method
tc	1125.88	K	Joback Method
tf	345.00 ± 0.20	K	NIST Webbook
vc	0.895	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	670.52	J/mol×K	891.34	Joback Method
cpg	681.22	J/mol×K	930.43	Joback Method
cpg	690.98	J/mol×K	969.52	Joback Method
cpg	699.87	J/mol×K	1008.61	Joback Method
cpg	707.97	J/mol×K	1047.70	Joback Method
cpg	715.37	J/mol×K	1086.79	Joback Method
cpg	722.15	J/mol×K	1125.88	Joback Method
dvisc	0.0002718	Paxs	539.47	Joback Method
dvisc	0.0001181	Paxs	598.12	Joback Method
dvisc	0.0000596	Paxs	656.76	Joback Method
dvisc	0.0000336	Paxs	715.40	Joback Method
dvisc	0.0000207	Paxs	774.05	Joback Method
dvisc	0.0000136	Paxs	832.69	Joback Method
dvisc	0.0000095	Paxs	891.34	Joback Method

## Sources

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**Aqueous Solubility Prediction Method:**

<http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa>

**Estimated Solubility Method:**

[http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl\\_file/ci034243xsi20040112\\_053635.txt](http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl_file/ci034243xsi20040112_053635.txt)

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C5836102&Units=SI>

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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