

# 5-Amino-2-methoxyphenol, N,N,O-tris(trifluoroacetyl)-

**Inchi:** InChI=1S/C13H6F9NO5/c1-27-6-3-2-5(4-7(6)28-10(26)13(20,21)22)23(8(24)11(14,15)16

**InchiKey:** JELDXANQGKOGEG-UHFFFAOYSA-N

**Formula:** C13H6F9NO5

**SMILES:** COc1ccc(N(C(=O)C(F)(F)F)C(=O)C(F)(F)F)cc1OC(=O)C(F)(F)F

**Mol. weight [g/mol]:** 427.18

## Physical Properties

Property code	Value	Unit	Source
gf	-2079.02	kJ/mol	Joback Method
hf	-2423.95	kJ/mol	Joback Method
hfus	38.36	kJ/mol	Joback Method
hvap	63.99	kJ/mol	Joback Method
log10ws	-4.19		Crippen Method
logp	3.147		Crippen Method
mcvol	212.630	ml/mol	McGowan Method
pc	1795.46	kPa	Joback Method
rinpol	1273.00		NIST Webbook
rinpol	1273.00		NIST Webbook
tb	736.11	K	Joback Method
tc	916.41	K	Joback Method
tf	527.02	K	Joback Method
vc	0.857	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	618.03	J/mol×K	736.11	Joback Method
cpg	627.12	J/mol×K	766.16	Joback Method
cpg	635.45	J/mol×K	796.21	Joback Method
cpg	643.08	J/mol×K	826.26	Joback Method
cpg	650.03	J/mol×K	856.31	Joback Method
cpg	656.37	J/mol×K	886.36	Joback Method
cpg	662.13	J/mol×K	916.41	Joback Method

# Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U374875&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U374875&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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