

# trans-2-Hexyl-5-methyl-1,3-dioxane

|                      |                                                                                  |
|----------------------|----------------------------------------------------------------------------------|
| Inchi:               | InChI=1S/C11H22O2/c1-3-4-5-6-7-11-12-8-10(2)9-13-11/h10-11H,3-9H2,1-2H3/t10-,11- |
| InchiKey:            | ZOVQWQKIOCKGED-XYPYZODXSA-N                                                      |
| Formula:             | C11H22O2                                                                         |
| SMILES:              | CCCCCCC1OCC(C)CO1                                                                |
| Mol. weight [g/mol]: | 186.29                                                                           |
| CAS:                 | 41824-31-1                                                                       |

## Physical Properties

| Property code | Value            | Unit    | Source         |
|---------------|------------------|---------|----------------|
| chl           | -6893.00 ± 13.00 | kJ/mol  | NIST Webbook   |
| gf            | -113.76          | kJ/mol  | Joback Method  |
| hf            | -500.39          | kJ/mol  | Joback Method  |
| hfl           | -580.00 ± 13.00  | kJ/mol  | NIST Webbook   |
| hfus          | 33.11            | kJ/mol  | Joback Method  |
| hvap          | 49.22            | kJ/mol  | Joback Method  |
| log10ws       | -2.86            |         | Crippen Method |
| logp          | 2.966            |         | Crippen Method |
| mcvol         | 166.730          | ml/mol  | McGowan Method |
| pc            | 2224.99          | kPa     | Joback Method  |
| tb            | 519.86           | K       | Joback Method  |
| tc            | 712.81           | K       | Joback Method  |
| tf            | 270.01           | K       | Joback Method  |
| vc            | 0.625            | m3/kmol | Joback Method  |

## Temperature Dependent Properties

| Property code | Value  | Unit    | Temperature [K] | Source        |
|---------------|--------|---------|-----------------|---------------|
| cpg           | 416.76 | J/mol×K | 519.86          | Joback Method |
| cpg           | 435.96 | J/mol×K | 552.02          | Joback Method |
| cpg           | 454.26 | J/mol×K | 584.18          | Joback Method |
| cpg           | 471.69 | J/mol×K | 616.34          | Joback Method |
| cpg           | 488.24 | J/mol×K | 648.49          | Joback Method |
| cpg           | 503.96 | J/mol×K | 680.65          | Joback Method |
| cpg           | 518.84 | J/mol×K | 712.81          | Joback Method |

|       |           |      |        |               |
|-------|-----------|------|--------|---------------|
| dvisc | 0.0055245 | Pa×s | 270.01 | Joback Method |
| dvisc | 0.0024098 | Pa×s | 311.65 | Joback Method |
| dvisc | 0.0012782 | Pa×s | 353.29 | Joback Method |
| dvisc | 0.0007750 | Pa×s | 394.94 | Joback Method |
| dvisc | 0.0005170 | Pa×s | 436.58 | Joback Method |
| dvisc | 0.0003700 | Pa×s | 478.22 | Joback Method |
| dvisc | 0.0002794 | Pa×s | 519.86 | Joback Method |

## Sources

|                        |                                                                                                                                               |
|------------------------|-----------------------------------------------------------------------------------------------------------------------------------------------|
| <b>NIST Webbook:</b>   | <a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C41824311&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C41824311&amp;Units=SI</a> |
| <b>Crippen Method:</b> | <a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>                                     |
| <b>Crippen Method:</b> | <a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>                             |
| <b>Joback Method:</b>  | <a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>                                         |
| <b>McGowan Method:</b> | <a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>                         |

## Legend

|                 |                                                           |
|-----------------|-----------------------------------------------------------|
| <b>chl:</b>     | Standard liquid enthalpy of combustion                    |
| <b>cpg:</b>     | Ideal gas heat capacity                                   |
| <b>dvisc:</b>   | Dynamic viscosity                                         |
| <b>gf:</b>      | Standard Gibbs free energy of formation                   |
| <b>hf:</b>      | Enthalpy of formation at standard conditions              |
| <b>hfl:</b>     | Liquid phase enthalpy of formation at standard conditions |
| <b>hfus:</b>    | Enthalpy of fusion at standard conditions                 |
| <b>hvap:</b>    | Enthalpy of vaporization at standard conditions           |
| <b>log10ws:</b> | Log10 of Water solubility in mol/l                        |
| <b>logp:</b>    | Octanol/Water partition coefficient                       |
| <b>mcvol:</b>   | McGowan's characteristic volume                           |
| <b>pc:</b>      | Critical Pressure                                         |
| <b>tb:</b>      | Normal Boiling Point Temperature                          |
| <b>tc:</b>      | Critical Temperature                                      |
| <b>tf:</b>      | Normal melting (fusion) point                             |
| <b>vc:</b>      | Critical Volume                                           |

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