

# Bicyclo[3.1.1]hept-2-en-4-ol, 2,6,6-trimethyl-, acetate

Other names:	verbenyl acetate verbenol acetate
Inchi:	InChI=1S/C12H18O2/c1-7-5-11(14-8(2)13)10-6-9(7)12(10,3)4/h5,9-11H,6H2,1-4H3
InchiKey:	OZBFUQLOVFXDNK-UHFFFAOYSA-N
Formula:	C12H18O2
SMILES:	CC(=O)OC1C=C(C)C2CC1C2(C)C
Mol. weight [g/mol]:	194.27

## Physical Properties

Property code	Value	Unit	Source
gf	-74.94	kJ/mol	Joback Method
hf	-375.50	kJ/mol	Joback Method
hfus	20.47	kJ/mol	Joback Method
hvap	50.64	kJ/mol	Joback Method
log10ws	-2.74		Crippen Method
logp	2.540		Crippen Method
mcvol	161.360	ml/mol	McGowan Method
pc	2405.28	kPa	Joback Method
rinpol	1297.00		NIST Webbook
rinpol	1294.00		NIST Webbook
rinpol	1290.00		NIST Webbook
rinpol	1282.00		NIST Webbook
rinpol	1293.00		NIST Webbook
rinpol	1292.00		NIST Webbook
rinpol	1293.00		NIST Webbook
rinpol	1321.00		NIST Webbook
rinpol	1294.00		NIST Webbook
rinpol	1294.00		NIST Webbook
rinpol	1287.00		NIST Webbook
rinpol	1293.00		NIST Webbook
rinpol	1298.00		NIST Webbook
rinpol	1293.00		NIST Webbook
rinpol	1290.00		NIST Webbook
rinpol	1282.00		NIST Webbook
rinpol	1282.00		NIST Webbook
ripol	1690.00		NIST Webbook
tb	563.04	K	Joback Method

tc	773.40	K	Joback Method
tf	358.22	K	Joback Method
vc	0.620	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	419.92	J/mol×K	563.04	Joback Method
cpg	437.32	J/mol×K	598.10	Joback Method
cpg	453.71	J/mol×K	633.16	Joback Method
cpg	469.20	J/mol×K	668.22	Joback Method
cpg	483.90	J/mol×K	703.28	Joback Method
cpg	497.94	J/mol×K	738.34	Joback Method
cpg	511.43	J/mol×K	773.40	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U195660&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U195660&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices

<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/32-334-5/Bicyclo-3-1-1-hept-2-en-4-ol-2-6-6-trimethyl-acetate.pdf>

Generated by Cheméo on 2024-04-17 02:23:08.926280188 +0000 UTC m=+15609837.846857509.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.