

# Phthalic acid, ethyl 3-phenoxybenzyl ester

<b>Inchi:</b>	InChI=1S/C23H20O5/c1-2-26-22(24)20-13-6-7-14-21(20)23(25)27-16-17-9-8-12-19(15-1
<b>InchiKey:</b>	UAFREMQYZFWXQP-UHFFFAOYSA-N
<b>Formula:</b>	C23H20O5
<b>SMILES:</b>	CCOC(=O)c1cccc1C(=O)OCc1cccc(Oc2ccccc2)c1
<b>Mol. weight [g/mol]:</b>	376.40

## Physical Properties

Property code	Value	Unit	Source
gf	-112.09	kJ/mol	Joback Method
hf	-453.22	kJ/mol	Joback Method
hfus	43.43	kJ/mol	Joback Method
hvap	95.67	kJ/mol	Joback Method
log10ws	-6.12		Crippen Method
logp	5.013		Crippen Method
mcvol	284.400	ml/mol	McGowan Method
pc	1753.60	kPa	Joback Method
rinsol	2949.00		NIST Webbook
rinsol	2949.00		NIST Webbook
tb	990.64	K	Joback Method
tc	1236.90	K	Joback Method
tf	619.82	K	Joback Method
vc	1.065	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	875.19	J/molxK	990.64	Joback Method
cpg	913.35	J/molxK	1195.86	Joback Method
cpg	908.82	J/molxK	1154.81	Joback Method
cpg	902.78	J/molxK	1113.77	Joback Method
cpg	895.19	J/molxK	1072.73	Joback Method
cpg	886.01	J/molxK	1031.68	Joback Method
cpg	916.43	J/molxK	1236.90	Joback Method
dvisc	0.0000290	Paxs	990.64	Joback Method

dvisc	0.0000363	Paxs	928.84	Joback Method
dvisc	0.0000468	Paxs	867.03	Joback Method
dvisc	0.0000629	Paxs	805.23	Joback Method
dvisc	0.0000887	Paxs	743.43	Joback Method
dvisc	0.0001333	Paxs	681.62	Joback Method
dvisc	0.0002170	Paxs	619.82	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U357032&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U357032&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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