

# Cyclopent-3-enyl-cyclohexane

<b>Inchi:</b>	InChI=1S/C11H18/c1-2-6-10(7-3-1)11-8-4-5-9-11/h4-5,10-11H,1-3,6-9H2
<b>InchiKey:</b>	CSIKMWYVIPFJES-UHFFFAOYSA-N
<b>Formula:</b>	C11H18
<b>SMILES:</b>	C1=CCC(C2CCCCC2)C1
<b>Mol. weight [g/mol]:</b>	150.26

## Physical Properties

Property code	Value	Unit	Source
gf	132.70	kJ/mol	Joback Method
hf	-97.79	kJ/mol	Joback Method
hfus	11.24	kJ/mol	Joback Method
hvap	41.06	kJ/mol	Joback Method
log10ws	-3.59		Crippen Method
logp	3.533		Crippen Method
mcvol	139.830	ml/mol	McGowan Method
pc	2912.39	kPa	Joback Method
rinpol	1170.00		NIST Webbook
rinpol	1204.00		NIST Webbook
tb	485.07	K	Joback Method
tc	714.76	K	Joback Method
tf	232.77	K	Joback Method
vc	0.511	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	317.77	J/molxK	485.07	Joback Method
cpg	340.81	J/molxK	523.35	Joback Method
cpg	362.39	J/molxK	561.63	Joback Method
cpg	382.56	J/molxK	599.92	Joback Method
cpg	401.39	J/molxK	638.20	Joback Method
cpg	418.94	J/molxK	676.48	Joback Method
cpg	435.25	J/molxK	714.76	Joback Method
dvisc	0.0057733	Paxs	232.77	Joback Method

dvisc	0.0024798	Paxs	274.82	Joback Method
dvisc	0.0013329	Paxs	316.87	Joback Method
dvisc	0.0008286	Paxs	358.92	Joback Method
dvisc	0.0005692	Paxs	400.97	Joback Method
dvisc	0.0004198	Paxs	443.02	Joback Method
dvisc	0.0003264	Paxs	485.07	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R136393&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R136393&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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