

# 4,5-Dipropyl-4,5-bis-(4-tert-butylphenyl)octane

<b>Inchi:</b>	InChI=1S/C34H54/c1-11-23-33(24-12-2,29-19-15-27(16-20-29)31(5,6)7)34(25-13-3,26-1
<b>InchiKey:</b>	JHALHDOCRWHJSQ-UHFFFAOYSA-N
<b>Formula:</b>	C34H54
<b>SMILES:</b>	CCCC(CCC)(c1ccc(C(C)(C)C)cc1)C(CCC)(CCC)c1ccc(C(C)(C)C)cc1
<b>Mol. weight [g/mol]:</b>	462.79
<b>CAS:</b>	85668-72-0

## Physical Properties

Property code	Value	Unit	Source
chs	-20686.00 ± 6.10	kJ/mol	NIST Webbook
gf	452.32	kJ/mol	Joback Method
hf	-213.00 ± 4.60	kJ/mol	NIST Webbook
hfs	-411.00 ± 6.10	kJ/mol	NIST Webbook
hfus	41.46	kJ/mol	Joback Method
hsub	198.00	kJ/mol	NIST Webbook
hsub	198.00	kJ/mol	NIST Webbook
hvap	91.97	kJ/mol	Joback Method
log10ws	-10.89		Crippen Method
logp	10.658		Crippen Method
mcvol	442.400	ml/mol	McGowan Method
pc	702.10	kPa	Joback Method
tb	1027.72	K	Joback Method
tc	1260.77	K	Joback Method
tf	419.00 ± 1.00	K	NIST Webbook
vc	1.679	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1662.74	J/mol×K	1221.93	Joback Method
cpg	1553.18	J/mol×K	1027.72	Joback Method
cpg	1576.17	J/mol×K	1066.56	Joback Method
cpg	1598.38	J/mol×K	1105.40	Joback Method
cpg	1620.05	J/mol×K	1144.25	Joback Method

cpg	1641.43	J/mol×K	1183.09	Joback Method
cpg	1684.23	J/mol×K	1260.77	Joback Method
cps	468.10	J/mol×K	393.00	NIST Webbook
cps	724.20	J/mol×K	298.00	NIST Webbook
dvisc	0.0000047	Paxs	1027.72	Joback Method
dvisc	0.0001720	Paxs	560.50	Joback Method
dvisc	0.0000655	Paxs	638.37	Joback Method
dvisc	0.0000307	Paxs	716.24	Joback Method
dvisc	0.0000167	Paxs	794.11	Joback Method
dvisc	0.0000102	Paxs	871.98	Joback Method
dvisc	0.0000067	Paxs	949.85	Joback Method
hfust	36.11	kJ/mol	452.00	NIST Webbook
hfust	40.58	kJ/mol	419.00	NIST Webbook
sfust	79.90	J/mol×K	452.00	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C85668720&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C85668720&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>cps:</b>	Solid phase heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hsub:</b>	Enthalpy of sublimation at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient

<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>sfust:</b>	Entropy of fusion at a given temperature
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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