

Formaldehyde, (2,4-dinitrophenyl)hydrazone

| | |
|-----------------------------|--|
| Other names: | Methanal 2,4-dinitrophenylhydrazone |
| Inchi: | InChI=1S/C7H6N4O4/c1-8-9-6-3-2-5(10(12)13)4-7(6)11(14)15/h2-4,9H,1H2 |
| InchiKey: | UEQLSLWCHGLSML-UHFFFAOYSA-N |
| Formula: | C7H6N4O4 |
| SMILES: | <chem>C=NNc1ccc([N+](=O)[O-])cc1[N+](=O)[O-]</chem> |
| Mol. weight [g/mol]: | 210.15 |
| CAS: | 1081-15-8 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|--------|----------------|
| hf | 148.16 | kJ/mol | Joback Method |
| hvap | 77.08 | kJ/mol | Joback Method |
| log10ws | -2.94 | | Crippen Method |
| logp | 1.530 | | Crippen Method |
| mcvol | 136.230 | ml/mol | McGowan Method |
| pc | 3598.56 | kPa | Joback Method |
| tb | 819.25 | K | Joback Method |
| tc | 1094.48 | K | Joback Method |

Sources

| | |
|------------------------|---|
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=C1081158&Units=SI |
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci990307I |
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |

Legend

| | |
|--------------|---|
| hf: | Enthalpy of formation at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |

| | |
|-----------------|-------------------------------------|
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |

Latest version available from:

<https://www.cheméo.com/cid/32-583-9/Formaldehyde-2-4-dinitrophenyl-hydrazone.pdf>

Generated by Cheméo on 2024-04-18 14:45:01.108684456 +0000 UTC m=+15740750.029261774.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.