

# Diethylmalonic acid, di(2-acetylphenyl) ester

<b>Inchi:</b>	InChI=1S/C23H24O6/c1-5-23(6-2,21(26)28-19-13-9-7-11-17(19)15(3)24)22(27)29-20-14
<b>InchiKey:</b>	ZCNJLPSKTIXTDT-UHFFFAOYSA-N
<b>Formula:</b>	C23H24O6
<b>SMILES:</b>	CCC(CC)(C(=O)Oc1ccccc1C(C)=O)C(=O)Oc1ccccc1C(C)=O
<b>Mol. weight [g/mol]:</b>	396.43

## Physical Properties

Property code	Value	Unit	Source
gf	-374.50	kJ/mol	Joback Method
hf	-791.44	kJ/mol	Joback Method
hfus	43.99	kJ/mol	Joback Method
hvap	103.18	kJ/mol	Joback Method
log10ws	-6.08		Crippen Method
logp	4.409		Crippen Method
mcvol	305.430	ml/mol	McGowan Method
pc	1529.46	kPa	Joback Method
rinpol	2838.00		NIST Webbook
rinpol	2838.00		NIST Webbook
tb	1046.05	K	Joback Method
tc	1289.12	K	Joback Method
tf	673.45	K	Joback Method
vc	1.157	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	970.54	J/mol×K	1046.05	Joback Method
cpg	980.62	J/mol×K	1086.56	Joback Method
cpg	989.36	J/mol×K	1127.07	Joback Method
cpg	996.81	J/mol×K	1167.58	Joback Method
cpg	1003.07	J/mol×K	1208.09	Joback Method
cpg	1008.21	J/mol×K	1248.60	Joback Method
cpg	1012.29	J/mol×K	1289.12	Joback Method
dvisc	0.0001993	Paxs	673.45	Joback Method

dvisc	0.0001221	Paxs	735.55	Joback Method
dvisc	0.0000807	Paxs	797.65	Joback Method
dvisc	0.0000566	Paxs	859.75	Joback Method
dvisc	0.0000417	Paxs	921.85	Joback Method
dvisc	0.0000319	Paxs	983.95	Joback Method
dvisc	0.0000252	Paxs	1046.05	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U370102&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U370102&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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