

Propane, 1-(ethylthio)-

Other names:	1-(Ethylthio)propane 3-Thiahexane Ethyl n-propyl sulfide Ethyl propyl sulfide Ethyl propyl sulphide Ethyl propyl thioether Propyl ethyl sulfide Sulfide, ethyl propyl n-C3H7SC2H5
Inchi:	InChI=1S/C5H12S/c1-3-5-6-4-2/h3-5H2,1-2H3
InchiKey:	ZDDDFDQTSXYYSE-UHFFFAOYSA-N
Formula:	C5H12S
SMILES:	CCCSCC
Mol. weight [g/mol]:	104.22
CAS:	4110-50-3

Physical Properties

Property code	Value	Unit	Source
chl	-4139.90 ± 0.59	kJ/mol	NIST Webbook
chl	-4139.50 ± 0.59	kJ/mol	NIST Webbook
hf	-104.50	kJ/mol	NIST Webbook
hf	-104.70 ± 0.79	kJ/mol	NIST Webbook
hfl	-144.60 ± 0.75	kJ/mol	NIST Webbook
hfl	-144.80 ± 0.79	kJ/mol	NIST Webbook
hfus	12.84	kJ/mol	Joback Method
hvap	40.08	kJ/mol	NIST Webbook
hvap	40.00	kJ/mol	NIST Webbook
hvap	39.50	kJ/mol	NIST Webbook
hvap	40.01	kJ/mol	NIST Webbook
hvap	40.10 ± 0.04	kJ/mol	NIST Webbook
hvap	40.10	kJ/mol	NIST Webbook
ie	8.37	eV	NIST Webbook
ie	8.50 ± 0.05	eV	NIST Webbook
log10ws	-2.16		Cheméo Relay (1.0)
logp	2.150		Crippen Method
mcvol	97.660	ml/mol	McGowan Method
pc	3538.87	kPa	Joback Method

rinpol	794.00		NIST Webbook
rinpol	794.00		NIST Webbook
rinpol	794.00		NIST Webbook
rinpol	785.00		NIST Webbook
rinpol	783.00		NIST Webbook
rinpol	782.00		NIST Webbook
rinpol	785.00		NIST Webbook
rinpol	785.00		NIST Webbook
rinpol	801.00		NIST Webbook
ripol	1017.00		NIST Webbook
ripol	1017.00		NIST Webbook
sl	309.53	J/mol×K	NIST Webbook
tb	391.70 ± 0.30	K	NIST Webbook
tb	389.00 ± 3.00	K	NIST Webbook
tb	391.70	K	NIST Webbook
tb	391.70	K	NIST Webbook
tc	573.62	K	Cheméo Relay (1.0)
tf	156.11 ± 0.10	K	NIST Webbook
tt	156.10 ± 0.04	K	NIST Webbook
tt	156.10 ± 0.03	K	NIST Webbook
tt	156.10 ± 0.04	K	NIST Webbook
tt	156.10 ± 0.03	K	NIST Webbook

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	169.71	J/mol×K	382.58	Joback Method
cpg	224.63	J/mol×K	572.69	Joback Method
cpg	216.36	J/mol×K	541.00	Joback Method
cpg	207.75	J/mol×K	509.32	Joback Method
cpg	198.78	J/mol×K	477.63	Joback Method
cpg	189.45	J/mol×K	445.95	Joback Method
cpg	179.76	J/mol×K	414.26	Joback Method
cpl	198.41	J/mol×K	298.15	NIST Webbook
hfust	10.58	kJ/mol	156.10	NIST Webbook
hfust	10.58	kJ/mol	156.10	NIST Webbook
hfust	10.58	kJ/mol	156.10	NIST Webbook
hfust	10.58	kJ/mol	156.10	NIST Webbook
hvapt	39.80	kJ/mol	355.50	NIST Webbook
hvapt	34.24	kJ/mol	391.70	NIST Webbook
hvapt	37.80	kJ/mol	364.50	NIST Webbook

sfust	67.78	J/mol×K	156.10	NIST Webbook
sfust	67.78	J/mol×K	156.10	NIST Webbook

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.42227e+01
Coeff. B	-3.21450e+03
Coeff. C	-5.70070e+01
Temperature range (K), min.	287.68
Temperature range (K), max.	417.73

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/T + C \cdot \ln(T) + D \cdot T^2$
Coeff. A	9.12345e+01
Coeff. B	-7.82974e+03
Coeff. C	-1.13601e+01
Coeff. D	7.81191e-06
Temperature range (K), min.	287.15
Temperature range (K), max.	584.00

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Cheméo Relay (1.0):	
KDB Vapor Pressure Data:	https://www.therc.org/research/kdb/hcprop/showprop.php?cmpid=1832
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C4110503&Units=SI
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Cheméo Relay (1.0, beta):	
The Yaws Handbook of Vapor Pressure:	https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure
KDB:	https://www.therc.org/files/research/kdb/mol/mol1832.mol
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

chl:	Standard liquid enthalpy of combustion
cpg:	Ideal gas heat capacity
cpl:	Liquid phase heat capacity
hf:	Enthalpy of formation at standard conditions
hfl:	Liquid phase enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hfust:	Enthalpy of fusion at a given temperature
hvap:	Enthalpy of vaporization at standard conditions
hvapt:	Enthalpy of vaporization at a given temperature
ie:	Ionization energy
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
pvap:	Vapor pressure
rinpol:	Non-polar retention indices
ripol:	Polar retention indices
sfust:	Entropy of fusion at a given temperature
sl:	Liquid phase molar entropy at standard conditions
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
tt:	Triple Point Temperature

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