

# Methanone, (2-amino-5-chlorophenyl)(2-chlorophenyl)-

Other names:	2-Amino-2',5-dichlorobenzophenone Benzophenone, 2-amino-2',5-dichloro- 2-Amino-5,2'-dichlor-benzophenone 2-Amino-5,2'-dichlorobenzophenone
Inchi:	InChI=1S/C13H9Cl2NO/c14-8-5-6-12(16)10(7-8)13(17)9-3-1-2-4-11(9)15/h1-7H,16H2
InchiKey:	KWZYIAJRFJVQDO-UHFFFAOYSA-N
Formula:	C13H9Cl2NO
SMILES:	<chem>Nc1ccc(Cl)cc1C(=O)c1ccccc1Cl</chem>
Mol. weight [g/mol]:	266.12
CAS:	2958-36-3

## Physical Properties

Property code	Value	Unit	Source
gf	168.18	kJ/mol	Joback Method
hf	16.73	kJ/mol	Joback Method
hfus	31.53	kJ/mol	Joback Method
hvap	77.23	kJ/mol	Joback Method
log10ws	-4.43		Crippen Method
logp	3.807		Crippen Method
mcvol	182.540	ml/mol	McGowan Method
pc	3076.16	kPa	Joback Method
rinpol	2145.00		NIST Webbook
rinpol	2191.00		NIST Webbook
rinpol	2195.00		NIST Webbook
rinpol	2195.00		NIST Webbook
rinpol	2145.00		NIST Webbook
tb	766.40	K	Joback Method
tc	1030.46	K	Joback Method
tf	519.70	K	Joback Method
vc	0.680	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
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cpg	438.38	J/mol×K	766.40	Joback Method
cpg	449.32	J/mol×K	810.41	Joback Method
cpg	459.21	J/mol×K	854.42	Joback Method
cpg	468.13	J/mol×K	898.43	Joback Method
cpg	476.14	J/mol×K	942.44	Joback Method
cpg	483.32	J/mol×K	986.45	Joback Method
cpg	489.73	J/mol×K	1030.46	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2958363&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2958363&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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