

# DIETHYLMALONIC ACID

|                      |                                                                               |
|----------------------|-------------------------------------------------------------------------------|
| Other names:         | Malonic acid, diethyl-<br>Diethylmalonic acid<br>3,3-Pentanedicarboxylic acid |
| Inchi:               | InChI=1S/C7H12O4/c1-3-7(4-2,5(8)9)6(10)11/h3-4H2,1-2H3,(H,8,9)(H,10,11)       |
| InchiKey:            | LTMRRSWNXVJMBA-UHFFFAOYSA-N                                                   |
| Formula:             | C7H12O4                                                                       |
| SMILES:              | CCC(CC)(C(=O)O)C(=O)O                                                         |
| Mol. weight [g/mol]: | 160.17                                                                        |

## Physical Properties

| Property code | Value   | Unit    | Source             |
|---------------|---------|---------|--------------------|
| af            | 0.7641  |         | Cheméo Relay (1.0) |
| gf            | -520.58 | kJ/mol  | Joback Method      |
| hf            | -858.63 | kJ/mol  | Cheméo Relay (1.0) |
| hfus          | 17.85   | kJ/mol  | Joback Method      |
| hvap          | 72.07   | kJ/mol  | Cheméo Relay (1.0) |
| ie            | 10.40   | eV      | NIST Webbook       |
| logp          | 0.962   |         | Crippen Method     |
| mcvol         | 124.370 | ml/mol  | McGowan Method     |
| pc            | 4103.88 | kPa     | Joback Method      |
| tb            | 500.24  | K       | Cheméo Relay (1.0) |
| tc            | 735.86  | K       | Cheméo Relay (1.0) |
| tf            | 405.87  | K       | Cheméo Relay (1.0) |
| vc            | 0.423   | m3/kmol | Cheméo Relay (1.0) |

## Temperature Dependent Properties

| Property code | Value  | Unit    | Temperature [K] | Source        |
|---------------|--------|---------|-----------------|---------------|
| cpg           | 327.90 | J/mol×K | 648.43          | Joback Method |
| cpg           | 335.90 | J/mol×K | 678.49          | Joback Method |
| cpg           | 343.44 | J/mol×K | 708.56          | Joback Method |
| cpg           | 350.53 | J/mol×K | 738.62          | Joback Method |
| cpg           | 357.22 | J/mol×K | 768.69          | Joback Method |
| cpg           | 363.51 | J/mol×K | 798.75          | Joback Method |

|       |           |         |        |               |
|-------|-----------|---------|--------|---------------|
| cpg   | 369.43    | J/mol×K | 828.81 | Joback Method |
| dvisc | 0.0046479 | Pa×s    | 392.57 | Joback Method |
| dvisc | 0.0012880 | Pa×s    | 435.21 | Joback Method |
| dvisc | 0.0004488 | Pa×s    | 477.86 | Joback Method |
| dvisc | 0.0001859 | Pa×s    | 520.50 | Joback Method |
| dvisc | 0.0000880 | Pa×s    | 563.14 | Joback Method |
| dvisc | 0.0000463 | Pa×s    | 605.79 | Joback Method |
| dvisc | 0.0000265 | Pa×s    | 648.43 | Joback Method |

## Sources

|                                  |                                                                                                                                           |
|----------------------------------|-------------------------------------------------------------------------------------------------------------------------------------------|
| <b>NIST Webbook:</b>             | <a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C510203&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C510203&amp;Units=SI</a> |
| <b>Joback Method:</b>            | <a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>                                     |
| <b>McGowan Method:</b>           | <a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>                     |
| <b>Crippen Method:</b>           | <a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>                                 |
| <b>Crippen Method:</b>           | <a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>                         |
| <b>Cheméo Relay (1.0):</b>       |                                                                                                                                           |
| <b>Cheméo Relay (1.0, beta):</b> |                                                                                                                                           |

## Legend

|               |                                                 |
|---------------|-------------------------------------------------|
| <b>af:</b>    | Acentric Factor                                 |
| <b>cpg:</b>   | Ideal gas heat capacity                         |
| <b>dvisc:</b> | Dynamic viscosity                               |
| <b>gf:</b>    | Standard Gibbs free energy of formation         |
| <b>hf:</b>    | Enthalpy of formation at standard conditions    |
| <b>hfus:</b>  | Enthalpy of fusion at standard conditions       |
| <b>hvap:</b>  | Enthalpy of vaporization at standard conditions |
| <b>ie:</b>    | Ionization energy                               |
| <b>logp:</b>  | Octanol/Water partition coefficient             |
| <b>mcvol:</b> | McGowan's characteristic volume                 |
| <b>pc:</b>    | Critical Pressure                               |
| <b>tb:</b>    | Normal Boiling Point Temperature                |
| <b>tc:</b>    | Critical Temperature                            |
| <b>tf:</b>    | Normal melting (fusion) point                   |
| <b>vc:</b>    | Critical Volume                                 |

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