

# 2-Cyclopenten-1-one, 2,3-dimethyl-

<b>Other names:</b>	2,3-dimethyl-2-cyclopentenone 2,3-Dimethylcyclopent-2-en-1-one Cyclopent-2-en-1-one, 2,3-dimethyl 2,3-Dimethyl-2-cyclopenten-1-one
<b>Inchi:</b>	InChI=1S/C7H10O/c1-5-3-4-7(8)6(5)2/h3-4H2,1-2H3
<b>InchiKey:</b>	UZLQSPYGTUMKGS-UHFFFAOYSA-N
<b>Formula:</b>	C7H10O
<b>SMILES:</b>	CC1=C(C)C(=O)CC1
<b>Mol. weight [g/mol]:</b>	110.15
<b>CAS:</b>	1121-05-7

## Physical Properties

Property code	Value	Unit	Source
gf	-59.57	kJ/mol	Joback Method
hf	-209.85	kJ/mol	Joback Method
hfus	6.70	kJ/mol	Joback Method
hvap	37.61	kJ/mol	Joback Method
log10ws	-1.78		Crippen Method
logp	1.686		Crippen Method
mcvol	95.900	ml/mol	McGowan Method
pc	3736.23	kPa	Joback Method
rinpol	999.00		NIST Webbook
rinpol	1001.00		NIST Webbook
rinpol	1043.00		NIST Webbook
rinpol	1035.00		NIST Webbook
rinpol	1052.00		NIST Webbook
rinpol	999.00		NIST Webbook
rinpol	1052.00		NIST Webbook
rinpol	1034.00		NIST Webbook
rinpol	1043.00		NIST Webbook
rinpol	1040.00		NIST Webbook
ripol	1523.00		NIST Webbook
ripol	1531.00		NIST Webbook
ripol	1530.00		NIST Webbook
ripol	1521.00		NIST Webbook
ripol	1521.00		NIST Webbook
ripol	1524.00		NIST Webbook

ripol	1535.00		NIST Webbook
ripol	1501.00		NIST Webbook
ripol	1524.00		NIST Webbook
ripol	1550.00		NIST Webbook
ripol	1573.00		NIST Webbook
ripol	1550.00		NIST Webbook
ripol	1582.00		NIST Webbook
ripol	1582.00		NIST Webbook
ripol	1535.00		NIST Webbook
ripol	1541.00		NIST Webbook
ripol	1535.00		NIST Webbook
tb	456.45	K	Joback Method
tc	676.77	K	Joback Method
tf	277.81	K	Joback Method
vc	0.362	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	190.88	J/mol×K	456.45	Joback Method
cpg	202.98	J/mol×K	493.17	Joback Method
cpg	214.61	J/mol×K	529.89	Joback Method
cpg	225.75	J/mol×K	566.61	Joback Method
cpg	236.40	J/mol×K	603.33	Joback Method
cpg	246.56	J/mol×K	640.05	Joback Method
cpg	256.22	J/mol×K	676.77	Joback Method

## Sources

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C1121057&Units=SI>

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>ripola:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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