

Isonovalal

Inchi:	InChI=1S/C10H16O/c1-8(2)10(4)6-5-9(3)7-11/h5,7-8H,4,6H2,1-3H3/b9-5-
InchiKey:	KDMDLRXLBMMBSU-UITAMQMPSA-N
Formula:	C10H16O
SMILES:	<chem>C=C(CC=C(C)C=O)C(C)C</chem>
Mol. weight [g/mol]:	152.23
CAS:	112164-19-9

Physical Properties

Property code	Value	Unit	Source
gf	82.32	kJ/mol	Joback Method
hf	-117.52	kJ/mol	Joback Method
hfus	16.72	kJ/mol	Joback Method
hvap	43.63	kJ/mol	Joback Method
log10ws	-2.75		Crippen Method
logp	2.734		Crippen Method
mcvol	144.730	ml/mol	McGowan Method
pc	2520.12	kPa	Joback Method
rinpol	1169.00		NIST Webbook
rinpol	1169.00		NIST Webbook
tb	477.02	K	Joback Method
tc	667.36	K	Joback Method
tf	194.70	K	Joback Method
vc	0.570	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	308.54	J/mol×K	477.02	Joback Method
cpg	322.55	J/mol×K	508.74	Joback Method
cpg	335.84	J/mol×K	540.47	Joback Method
cpg	348.43	J/mol×K	572.19	Joback Method
cpg	360.36	J/mol×K	603.91	Joback Method
cpg	371.65	J/mol×K	635.63	Joback Method
cpg	382.35	J/mol×K	667.36	Joback Method

Sources

Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C112164199&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
h vap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
r in pol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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