

# Pentane, 3-ethyl-2,2,3-trimethyl-

<b>Other names:</b>	2,2,3-Trimethyl-3-ethylpentane 3-Ethyl-2,2,3-trimethylpentane
<b>Inchi:</b>	InChI=1S/C10H22/c1-7-10(6,8-2)9(3,4)5/h7-8H2,1-6H3
<b>InchiKey:</b>	AJDIFHIHSYVDGP-UHFFFAOYSA-N
<b>Formula:</b>	C10H22
<b>SMILES:</b>	CCC(C)(CC)C(C)(C)C
<b>Mol. weight [g/mol]:</b>	142.28
<b>CAS:</b>	52897-17-3

## Physical Properties

Property code	Value	Unit	Source
gf	39.00	kJ/mol	Joback Method
hf	-267.23	kJ/mol	Joback Method
hfus	6.83	kJ/mol	Joback Method
hvap	46.00	kJ/mol	NIST Webbook
log10ws	-3.52		Crippen Method
logp	3.859		Crippen Method
mcvol	151.760	ml/mol	McGowan Method
pc	2183.60	kPa	Joback Method
rinpol	966.00		NIST Webbook
rinpol	967.30		NIST Webbook
rinpol	966.00		NIST Webbook
rinpol	966.00		NIST Webbook
rinpol	965.70		NIST Webbook
rinpol	966.00		NIST Webbook
tb	421.74	K	Joback Method
tc	605.42	K	Joback Method
tf	230.90 ± 2.00	K	NIST Webbook
vc	0.574	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	318.50	J/molxK	421.74	Joback Method

cpg	336.71	J/molxK	452.35	Joback Method
cpg	353.94	J/molxK	482.97	Joback Method
cpg	370.23	J/molxK	513.58	Joback Method
cpg	385.63	J/molxK	544.19	Joback Method
cpg	400.18	J/molxK	574.81	Joback Method
cpg	413.92	J/molxK	605.42	Joback Method
dvisc	0.0186536	Paxs	207.30	Joback Method
dvisc	0.0054554	Paxs	243.04	Joback Method
dvisc	0.0021867	Paxs	278.78	Joback Method
dvisc	0.0010789	Paxs	314.52	Joback Method
dvisc	0.0006149	Paxs	350.26	Joback Method
dvisc	0.0003889	Paxs	386.00	Joback Method
dvisc	0.0002658	Paxs	421.74	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.40807e+01
Coeff. B	-3.70401e+03
Coeff. C	-5.12120e+01
Temperature range (K), min.	319.75
Temperature range (K), max.	473.60

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>KDB:</b>	<a href="https://www.thermo.com/files/research/kdb/mol/mol167.mol">https://www.thermo.com/files/research/kdb/mol/mol167.mol</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C52897173&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C52897173&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpolar:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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