

# Phenol, 4,4'-methylenebis-

<b>Other names:</b>	Phenol, 4,4'-methylenedi- p-(p-Hydroxybenzyl)phenol Bis(p-hydroxyphenyl)methane Bis(4-hydroxyphenyl)methane HDM 4,4'-Dihydroxydiphenylmethane 4,4'-Methylenebis[phenol] 4,4'-Methylenediphenol p,p'-Methylenediphenol p,p'-Bis(hydroxyphenyl)methane 4,4-Dihydroxydiphenylmethane Bisphenol F NSC 401136 4,4'-Bis(hydroxyphenyl)methane
<b>Inchi:</b>	InChI=1S/C13H12O2/c14-12-5-1-10(2-6-12)9-11-3-7-13(15)8-4-11/h1-8,14-15H,9H2
<b>InchiKey:</b>	PXKLMJQFEQBVL-D-UHFFFAOYSA-N
<b>Formula:</b>	C13H12O2
<b>SMILES:</b>	Oc1ccc(Cc2ccc(O)cc2)cc1
<b>Mol. weight [g/mol]:</b>	200.23
<b>CAS:</b>	620-92-8

## Physical Properties

Property code	Value	Unit	Source
gf	-25.84	kJ/mol	Joback Method
hf	-193.21	kJ/mol	Joback Method
hfus	29.07	kJ/mol	Joback Method
hvap	75.11	kJ/mol	Joback Method
log10ws	-2.67		Crippen Method
logp	2.689		Crippen Method
mcvol	158.250	ml/mol	McGowan Method
pc	4409.10	kPa	Joback Method
rinp	2065.00		NIST Webbook
tb	711.44	K	Joback Method
tc	973.14	K	Joback Method
tf	512.55	K	Joback Method
vc	0.479	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	423.36	J/molxK	711.44	Joback Method
cpg	480.26	J/molxK	929.53	Joback Method
cpg	469.60	J/molxK	885.91	Joback Method
cpg	458.81	J/molxK	842.29	Joback Method
cpg	447.66	J/molxK	798.67	Joback Method
cpg	435.92	J/molxK	755.06	Joback Method
cpg	491.03	J/molxK	973.14	Joback Method
dvisc	0.0000015	Paxs	711.44	Joback Method
dvisc	0.0000024	Paxs	678.29	Joback Method
dvisc	0.0000040	Paxs	645.14	Joback Method
dvisc	0.0000070	Paxs	612.00	Joback Method
dvisc	0.0000130	Paxs	578.85	Joback Method
dvisc	0.0000261	Paxs	545.70	Joback Method
dvisc	0.0000572	Paxs	512.55	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C620928&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C620928&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions

<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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