

# Behenyl chloride

<b>Other names:</b>	1-Chlorodocosane Docosane, 1-chloro-
<b>Inchi:</b>	InChI=1S/C22H45Cl/c1-2-3-4-5-6-7-8-9-10-11-12-13-14-15-16-17-18-19-20-21-22-23/h2
<b>InchiKey:</b>	OACXFSZVCDOBKF-UHFFFAOYSA-N
<b>Formula:</b>	C22H45Cl
<b>SMILES:</b>	CCCCCCCCCCCCCCCCCCCCCCCCI
<b>Mol. weight [g/mol]:</b>	345.05
<b>CAS:</b>	42217-03-8

## Physical Properties

Property code	Value	Unit	Source
gf	122.43	kJ/mol	Joback Method
hf	-513.15	kJ/mol	Joback Method
hfus	56.93	kJ/mol	Joback Method
hvap	68.95	kJ/mol	Joback Method
log10ws	-9.18		Crippen Method
logp	9.047		Crippen Method
mvol	333.080	ml/mol	McGowan Method
pc	882.09	kPa	Joback Method
ripol	2692.00		NIST Webbook
tb	740.19	K	Joback Method
tc	910.35	K	Joback Method
tf	367.62	K	Joback Method
vc	1.317	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1002.07	J/molxK	740.19	Joback Method
cpg	1023.16	J/molxK	768.55	Joback Method
cpg	1043.28	J/molxK	796.91	Joback Method
cpg	1062.48	J/molxK	825.27	Joback Method
cpg	1080.80	J/molxK	853.63	Joback Method
cpg	1098.27	J/molxK	881.99	Joback Method

cpg	1114.92	J/molxK	910.35	Joback Method
dvisc	0.0021878	Paxs	367.62	Joback Method
dvisc	0.0008019	Paxs	429.71	Joback Method
dvisc	0.0003787	Paxs	491.81	Joback Method
dvisc	0.0002116	Paxs	553.90	Joback Method
dvisc	0.0001330	Paxs	616.00	Joback Method
dvisc	0.0000910	Paxs	678.10	Joback Method
dvisc	0.0000663	Paxs	740.19	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.32945e+01
Coeff. B	-5.05479e+03
Coeff. C	-1.26896e+02
Temperature range (K), min.	515.52
Temperature range (K), max.	760.09

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C42217038&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C42217038&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions

<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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