

«beta»-Alanine, N-acryloyl-, butyl ester

Inchi:	InChI=1S/C10H17NO3/c1-3-5-8-14-10(13)6-7-11-9(12)4-2/h4H,2-3,5-8H2,1H3,(H,11,12)
InchiKey:	JPICICRJSFKZQY-UHFFFAOYSA-N
Formula:	C10H17NO3
SMILES:	C=CC(=O)NCCC(=O)OCCCC
Mol. weight [g/mol]:	199.25

Physical Properties

Property code	Value	Unit	Source
gf	-152.29	kJ/mol	Joback Method
hf	-428.21	kJ/mol	Joback Method
hfus	29.86	kJ/mol	Joback Method
hvap	59.52	kJ/mol	Joback Method
log10ws	-1.69		Crippen Method
logp	1.022		Crippen Method
mcvol	166.450	ml/mol	McGowan Method
pc	2465.36	kPa	Joback Method
rinpol	1578.00		NIST Webbook
rinpol	1578.00		NIST Webbook
tb	605.21	K	Joback Method
tc	791.03	K	Joback Method
tf	375.45	K	Joback Method
vc	0.641	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	420.50	J/mol×K	605.21	Joback Method
cpg	433.30	J/mol×K	636.18	Joback Method
cpg	445.47	J/mol×K	667.15	Joback Method
cpg	457.04	J/mol×K	698.12	Joback Method
cpg	468.00	J/mol×K	729.09	Joback Method
cpg	478.38	J/mol×K	760.06	Joback Method
cpg	488.19	J/mol×K	791.03	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U321677&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpola:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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