

# 10-Chlorodecyl ethyl carbonate

<b>Inchi:</b>	InChI=1S/C13H25ClO3/c1-2-16-13(15)17-12-10-8-6-4-3-5-7-9-11-14/h2-12H2,1H3
<b>InchiKey:</b>	RCJVEQQSPNMRBB-UHFFFAOYSA-N
<b>Formula:</b>	C13H25ClO3
<b>SMILES:</b>	CCOC(=O)OCCCCCCCCCCI
<b>Mol. weight [g/mol]:</b>	264.79

## Physical Properties

Property code	Value	Unit	Source
gf	-292.27	kJ/mol	Joback Method
hf	-704.41	kJ/mol	Joback Method
hfus	37.60	kJ/mol	Joback Method
hvap	60.48	kJ/mol	Joback Method
log10ws	-4.34		Crippen Method
logp	4.519		Crippen Method
mvol	219.580	ml/mol	McGowan Method
pc	1633.81	kPa	Joback Method
rmpol	1856.00		NIST Webbook
rmpol	1856.00		NIST Webbook
tb	632.98	K	Joback Method
tc	806.36	K	Joback Method
tf	360.58	K	Joback Method
vc	0.855	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	576.08	J/molxK	632.98	Joback Method
cpg	591.52	J/molxK	661.88	Joback Method
cpg	606.29	J/molxK	690.77	Joback Method
cpg	620.41	J/molxK	719.67	Joback Method
cpg	633.88	J/molxK	748.57	Joback Method
cpg	646.70	J/molxK	777.47	Joback Method
cpg	658.88	J/molxK	806.36	Joback Method
dvisc	0.0017014	Paxs	360.58	Joback Method

dvisc	0.0008610	Paxs	405.98	Joback Method
dvisc	0.0004997	Paxs	451.38	Joback Method
dvisc	0.0003203	Paxs	496.78	Joback Method
dvisc	0.0002212	Paxs	542.18	Joback Method
dvisc	0.0001618	Paxs	587.58	Joback Method
dvisc	0.0001237	Paxs	632.98	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U373787&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U373787&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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