

# Cyclopentane, 1-methyl-3-methylene

<b>Inchi:</b>	InChI=1S/C7H12/c1-6-3-4-7(2)5-6/h7H,1,3-5H2,2H3
<b>InchiKey:</b>	BMLMXLRJGCBOOF-UHFFFAOYSA-N
<b>Formula:</b>	C7H12
<b>SMILES:</b>	C=C1CCC(C)C1
<b>Mol. weight [g/mol]:</b>	96.17

## Physical Properties

Property code	Value	Unit	Source
gf	97.69	kJ/mol	Joback Method
hf	-43.09	kJ/mol	Joback Method
hfus	6.66	kJ/mol	Joback Method
hvap	31.59	kJ/mol	Joback Method
log10ws	-2.26		Crippen Method
logp	2.363		Crippen Method
mcvol	94.330	ml/mol	McGowan Method
pc	3497.14	kPa	Joback Method
rinpola	713.00		NIST Webbook
rinpola	706.00		NIST Webbook
rinpola	709.00		NIST Webbook
rinpola	709.00		NIST Webbook
tb	374.00	K	Joback Method
tc	569.84	K	Joback Method
tf	193.23	K	Joback Method
vc	0.352	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	163.26	J/molxK	374.00	Joback Method
cpg	176.80	J/molxK	406.64	Joback Method
cpg	189.72	J/molxK	439.28	Joback Method
cpg	202.03	J/molxK	471.92	Joback Method
cpg	213.76	J/molxK	504.56	Joback Method
cpg	224.91	J/molxK	537.20	Joback Method

cpg	235.52	J/mol×K	569.84	Joback Method
dvisc	0.0017344	Paxs	193.23	Joback Method
dvisc	0.0010433	Paxs	223.36	Joback Method
dvisc	0.0007082	Paxs	253.49	Joback Method
dvisc	0.0005220	Paxs	283.62	Joback Method
dvisc	0.0004080	Paxs	313.74	Joback Method
dvisc	0.0003329	Paxs	343.87	Joback Method
dvisc	0.0002807	Paxs	374.00	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R10678&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R10678&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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