

# Diethylmalonic acid, hexyl 2-methylthiophenyl ester

<b>Inchi:</b>	InChI=1S/C20H30O4S/c1-5-8-9-12-15-23-18(21)20(6-2,7-3)19(22)24-16-13-10-11-14-17
<b>InchiKey:</b>	HFVMHNPBSSOKQW-UHFFFAOYSA-N
<b>Formula:</b>	C20H30O4S
<b>SMILES:</b>	CCCCCOC(=O)C(CC)(CC)C(=O)Oc1cccc1SC
<b>Mol. weight [g/mol]:</b>	366.51

## Physical Properties

Property code	Value	Unit	Source
gf	-211.58	kJ/mol	Joback Method
hf	-687.55	kJ/mol	Joback Method
hfus	43.50	kJ/mol	Joback Method
hvap	86.88	kJ/mol	Joback Method
log10ws	-5.75		Crippen Method
logp	5.244		Crippen Method
mvol	300.130	ml/mol	McGowan Method
pc	1377.86	kPa	Joback Method
rinpol	2459.00		NIST Webbook
rinpol	2459.00		NIST Webbook
tb	906.79	K	Joback Method
tc	1125.12	K	Joback Method
tf	535.24	K	Joback Method
vc	1.139	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	937.34	J/mol×K	906.79	Joback Method
cpg	952.15	J/mol×K	943.18	Joback Method
cpg	965.65	J/mol×K	979.57	Joback Method
cpg	977.89	J/mol×K	1015.96	Joback Method
cpg	988.91	J/mol×K	1052.34	Joback Method
cpg	998.74	J/mol×K	1088.73	Joback Method
cpg	1007.44	J/mol×K	1125.12	Joback Method

# Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U369533&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U369533&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h vap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>r in pol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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