

# Pentachlorothioanisole

<b>Other names:</b>	Benzene, pentachloro(methylthio)- Methyl pentachlorophenyl sulfide Methylthiopentachlorobenzene Pentachloro(methylthio)benzene Pentachlorophenyl methyl sulfide Sulfide, methyl pentachlorophenyl Pctas 1,2,3,4,5-Pentachloro-6-(methylsulfanyl)benzene
<b>Inchi:</b>	InChI=1S/C7H3Cl5S/c1-13-7-5(11)3(9)2(8)4(10)6(7)12/h1H3
<b>InchiKey:</b>	LGZZJTIUEJNNKV-UHFFFAOYSA-N
<b>Formula:</b>	C7H3Cl5S
<b>SMILES:</b>	CSc1c(Cl)c(Cl)c(Cl)c(Cl)c1Cl
<b>Mol. weight [g/mol]:</b>	296.43
<b>CAS:</b>	1825-19-0

## Physical Properties

Property code	Value	Unit	Source
gf	45.79	kJ/mol	Joback Method
hf	-45.46	kJ/mol	Joback Method
hfus	31.10	kJ/mol	Joback Method
hvap	65.50	kJ/mol	Joback Method
log10ws	-5.64		Crippen Method
logp	5.675		Crippen Method
mcvol	163.280	ml/mol	McGowan Method
pc	3079.57	kPa	Joback Method
tb	667.07	K	Joback Method
tc	932.24	K	Joback Method
tf	441.67	K	Joback Method
vc	0.619	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	278.27	J/mol×K	667.07	Joback Method

cpg	284.86	J/mol×K	711.27	Joback Method
cpg	290.91	J/mol×K	755.46	Joback Method
cpg	296.44	J/mol×K	799.66	Joback Method
cpg	301.43	J/mol×K	843.85	Joback Method
cpg	305.88	J/mol×K	888.05	Joback Method
cpg	309.78	J/mol×K	932.24	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1825190&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1825190&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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