

# Propanoic acid, 2-methyl-, 2-methylpropyl ester

<b>Other names:</b>	2-METHYLPROPYL ISOBUTYRATE 2-Methylpropyl 2-methylpropanoate 2-Methylpropyl 2-methylpropionate 2-methylpropionic acid isobutyl ester ISOBUTYL ISOBUTANOATE Isobutyl 2-methylpropanoate Isobutyl ester of 2-methylpropanoic acid Isobutyl isobutyrate Isobutylester kyseliny isomaselne Isobutyric acid, isobutyl ester NSC 6538 UN 2528
<b>Inchi:</b>	InChI=1S/C8H16O2/c1-6(2)5-10-8(9)7(3)4/h6-7H,5H2,1-4H3
<b>InchiKey:</b>	RXGUIWHIADMCFE-UHFFFAOYSA-N
<b>Formula:</b>	C8H16O2
<b>SMILES:</b>	CC(C)COC(=O)C(C)C
<b>Mol. weight [g/mol]:</b>	144.21
<b>CAS:</b>	97-85-8

## Physical Properties

Property code	Value	Unit	Source
chl	-4847.40 ± 1.60	kJ/mol	NIST Webbook
gf	-222.32	kJ/mol	Joback Method
hf	-542.90 ± 1.70	kJ/mol	NIST Webbook
hfl	-587.40 ± 1.60	kJ/mol	NIST Webbook
hfus	12.22	kJ/mol	Joback Method
hvac	44.50 ± 0.10	kJ/mol	NIST Webbook
hvac	44.50	kJ/mol	NIST Webbook
hvac	48.50	kJ/mol	NIST Webbook
hvac	44.49 ± 0.39	kJ/mol	NIST Webbook
log10ws	-1.55		Crippen Method
logp	1.842		Crippen Method
mccol	131.020	ml/mol	McGowan Method
pc	2460.00	kPa	KDB
rinpol	895.00		NIST Webbook
rinpol	914.00		NIST Webbook
rinpol	900.00		NIST Webbook

rinpol	900.00	NIST Webbook
rinpol	925.00	NIST Webbook
rinpol	899.00	NIST Webbook
rinpol	900.00	NIST Webbook
rinpol	916.00	NIST Webbook
rinpol	906.00	NIST Webbook
rinpol	938.00	NIST Webbook
rinpol	900.00	NIST Webbook
rinpol	913.00	NIST Webbook
rinpol	900.00	NIST Webbook
rinpol	909.00	NIST Webbook
rinpol	892.00	NIST Webbook
rinpol	904.00	NIST Webbook
rinpol	909.00	NIST Webbook
rinpol	914.00	NIST Webbook
rinpol	905.00	NIST Webbook
rinpol	899.00	NIST Webbook
rinpol	901.00	NIST Webbook
rinpol	906.00	NIST Webbook
rinpol	913.80	NIST Webbook
rinpol	901.00	NIST Webbook
rinpol	906.30	NIST Webbook
rinpol	899.00	NIST Webbook
rinpol	900.00	NIST Webbook
rinpol	925.00	NIST Webbook
rinpol	899.00	NIST Webbook
rinpol	902.00	NIST Webbook
rinpol	899.00	NIST Webbook
rinpol	902.00	NIST Webbook
rinpol	906.00	NIST Webbook
rinpol	900.00	NIST Webbook
rinpol	899.00	NIST Webbook
rinpol	896.00	NIST Webbook
rinpol	900.00	NIST Webbook
rinpol	900.00	NIST Webbook
rinpol	909.00	NIST Webbook
ripol	1095.00	NIST Webbook
ripol	1092.00	NIST Webbook
ripol	1094.00	NIST Webbook
ripol	1091.00	NIST Webbook
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ripol	1085.00		NIST Webbook
ripol	1090.00		NIST Webbook
ripol	1103.50		NIST Webbook
ripol	1094.00		NIST Webbook
ripol	1083.00		NIST Webbook
ripol	1081.00		NIST Webbook
ripol	1090.00		NIST Webbook
tb	420.70	K	NIST Webbook
tb	421.80	K	KDB
tb	422.20 ± 2.00	K	NIST Webbook
tb	420.65 ± 1.50	K	NIST Webbook
tb	421.80 ± 0.25	K	NIST Webbook
tb	420.50 ± 0.50	K	NIST Webbook
tb	420.50 ± 0.50	K	NIST Webbook
tb	420.15 ± 1.50	K	NIST Webbook
tc	602.00	K	KDB
tc	601.89 ± 6.00	K	NIST Webbook
tf	192.40	K	KDB
tf	192.50 ± 0.15	K	NIST Webbook
vc	0.495	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	350.42	J/mol×K	640.65	Joback Method
cpg	292.89	J/mol×K	488.32	Joback Method
cpg	305.35	J/mol×K	518.78	Joback Method
cpg	317.32	J/mol×K	549.25	Joback Method
cpg	328.83	J/mol×K	579.72	Joback Method
cpg	339.86	J/mol×K	610.19	Joback Method
cpg	279.95	J/mol×K	457.85	Joback Method
dvisc	0.0003141	Paxs	418.56	Joback Method
dvisc	0.0004562	Paxs	379.26	Joback Method
dvisc	0.0007223	Paxs	339.97	Joback Method

dvisc	0.0012896	Paxs	300.67	Joback Method
dvisc	0.0027406	Paxs	261.38	Joback Method
dvisc	0.0002306	Paxs	457.85	Joback Method
dvisc	0.0076048	Paxs	222.08	Joback Method
hvapt	46.90	kJ/mol	349.00	NIST Webbook
pvap	0.60	kPa	298.20	Transpiration method: Vapor pressures and enthalpies of vaporization of some low-boiling esters
pvap	1.14	kPa	308.40	Transpiration method: Vapor pressures and enthalpies of vaporization of some low-boiling esters
pvap	1.36	kPa	311.20	Transpiration method: Vapor pressures and enthalpies of vaporization of some low-boiling esters
pvap	1.50	kPa	313.40	Transpiration method: Vapor pressures and enthalpies of vaporization of some low-boiling esters
pvap	1.85	kPa	316.20	Transpiration method: Vapor pressures and enthalpies of vaporization of some low-boiling esters
pvap	2.22	kPa	319.20	Transpiration method: Vapor pressures and enthalpies of vaporization of some low-boiling esters
pvap	0.43	kPa	293.20	Transpiration method: Vapor pressures and enthalpies of vaporization of some low-boiling esters
pvap	0.30	kPa	288.20	Transpiration method: Vapor pressures and enthalpies of vaporization of some low-boiling esters

pvap	0.21	kPa	283.20	Transpiration method: Vapor pressures and enthalpies of vaporization of some low-boiling esters
pvap	0.15	kPa	278.20	Transpiration method: Vapor pressures and enthalpies of vaporization of some low-boiling esters
pvap	0.13	kPa	276.20	Transpiration method: Vapor pressures and enthalpies of vaporization of some low-boiling esters
pvap	0.11	kPa	274.20	Transpiration method: Vapor pressures and enthalpies of vaporization of some low-boiling esters
pvap	0.83	kPa	303.30	Transpiration method: Vapor pressures and enthalpies of vaporization of some low-boiling esters
rho1	875.00	kg/m3	273.00	KDB

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/T + C \cdot \ln(T) + D \cdot T^2$
Coeff. A	7.73503e+01
Coeff. B	-8.03882e+03
Coeff. C	-9.01363e+00
Coeff. D	4.18948e-06
Temperature range (K), min.	192.45
Temperature range (K), max.	602.00

# Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C97858&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C97858&amp;Units=SI</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Transpiration method: Vapor pressures and enthalpies of vaporization of some low boiling esters.</b>	<a href="https://www.doi.org/10.1016/j.fluid.2008.02.001">https://www.doi.org/10.1016/j.fluid.2008.02.001</a>
<b>Equilibrium for the system of ionic liquid [BMim][HSO<sub>4</sub>] catalysed C<sub>10</sub>H<sub>8</sub>O<sub>2</sub> formation:</b>	<a href="https://www.doi.org/10.1016/j.jct.2018.03.015">https://www.doi.org/10.1016/j.jct.2018.03.015</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>KDB Vapor Pressure Data:</b>	<a href="https://www.cheric.org/research/kdb/hcprop/showprop.php?cmpid=1104">https://www.cheric.org/research/kdb/hcprop/showprop.php?cmpid=1104</a>
<b>KDB:</b>	<a href="https://www.cheric.org/files/research/kdb/mol/mol1104.mol">https://www.cheric.org/files/research/kdb/mol/mol1104.mol</a>

# Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rho:</b>	Liquid Density
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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