

# 3,7-Benzofurandiol, 2,3-dihydro-2,2-dimethyl-

<b>Other names:</b>	3-Hydroxycarbofuran phenol Carbofuran-3-hydroxy-7-phenol Carbofuran 7-phenol 3-oh 2,2-Dimethyl-2,3-dihydro-1-benzofuran-3,7-diol 2,2-Dimethyl-3,7-dihydroxy-2,3-dihydrobenzofuran 3-Hydroxycarbofuran-7-ol
<b>Inchi:</b>	InChI=1S/C10H12O3/c1-10(2)9(12)6-4-3-5-7(11)8(6)13-10/h3-5,9,11-12H,1-2H3
<b>InchiKey:</b>	CNFFSKAZCVPTPA-UHFFFAOYSA-N
<b>Formula:</b>	C10H12O3
<b>SMILES:</b>	CC1(C)Oc2c(O)cccc2C1O
<b>Mol. weight [g/mol]:</b>	180.20
<b>CAS:</b>	17781-15-6

## Physical Properties

Property code	Value	Unit	Source
gf	-193.91	kJ/mol	Joback Method
hf	-418.51	kJ/mol	Joback Method
hfus	26.07	kJ/mol	Joback Method
hvap	73.45	kJ/mol	Joback Method
log10ws	-2.09		Crippen Method
logp	1.597		Crippen Method
mcvol	134.750	ml/mol	McGowan Method
pc	4516.42	kPa	Joback Method
tb	661.92	K	Joback Method
tc	885.04	K	Joback Method
tf	383.94 ± 0.20	K	NIST Webbook
vc	0.448	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	371.83	J/mol×K	661.92	Joback Method
cpg	382.53	J/mol×K	699.11	Joback Method
cpg	392.75	J/mol×K	736.29	Joback Method

cpg	402.68	J/mol×K	773.48	Joback Method
cpg	412.52	J/mol×K	810.67	Joback Method
cpg	422.45	J/mol×K	847.85	Joback Method
cpg	432.65	J/mol×K	885.04	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C17781156&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C17781156&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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