

# Cyclohexane,1,2-dibromo-,trans-

<b>Other names:</b>	.+/--trans-1,2-Dibromocyclohexane trans-1,2-Dibromocyclohexane
<b>Inchi:</b>	InChI=1S/C6H10Br2/c7-5-3-1-2-4-6(5)8/h5-6H,1-4H2/t5-,6-/m0/s1
<b>InchiKey:</b>	CZNHKZKWKJNOTE-WDSKDSINSA-N
<b>Formula:</b>	C6H10Br2
<b>SMILES:</b>	BrC1CCCCC1Br
<b>Mol. weight [g/mol]:</b>	241.95
<b>CAS:</b>	7429-37-0

## Physical Properties

Property code	Value	Unit	Source
gf	45.02	kJ/mol	Joback Method
hf	-80.53	kJ/mol	Joback Method
hfus	14.77	kJ/mol	Joback Method
hvap	50.50	kJ/mol	NIST Webbook
ie	10.06 ± 0.01	eV	NIST Webbook
ie	10.02 ± 0.02	eV	NIST Webbook
log10ws	-3.32		Crippen Method
logp	3.087		Crippen Method
mcvol	119.540	ml/mol	McGowan Method
pc	4540.80	kPa	Joback Method
tb	483.88	K	Joback Method
tc	729.25	K	Joback Method
tf	280.12	K	Joback Method
vc	0.427	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	217.19	J/molxK	483.88	Joback Method
cpg	231.69	J/molxK	524.77	Joback Method
cpg	245.19	J/molxK	565.67	Joback Method
cpg	257.74	J/molxK	606.56	Joback Method
cpg	269.38	J/molxK	647.46	Joback Method

cpg	280.16	J/molxK	688.35	Joback Method
cpg	290.13	J/molxK	729.25	Joback Method
dvisc	0.0031464	Paxs	280.12	Joback Method
dvisc	0.0018892	Paxs	314.08	Joback Method
dvisc	0.0012531	Paxs	348.04	Joback Method
dvisc	0.0008941	Paxs	382.00	Joback Method
dvisc	0.0006741	Paxs	415.96	Joback Method
dvisc	0.0005303	Paxs	449.92	Joback Method
dvisc	0.0004316	Paxs	483.88	Joback Method
hvapt	53.30	kJ/mol	383.00	NIST Webbook

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	418.20	K	13.30	NIST Webbook
tbrp	374.00 ± 2.00	K	2.10	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.96449e+01
Coeff. B	-7.65778e+03
Coeff. C	3.17290e+01
Temperature range (K), min.	363.87
Temperature range (K), max.	502.53

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C7429370&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C7429370&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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