

# 1,1'-Biphenyl, 3,3'-dimethoxy-

<b>Other names:</b>	Biphenyl, 3,3'-dimethoxy- 3,3'-Dimethoxybiphenyl 3,3'-dimethoxy-1,1'-biphenyl
<b>Inchi:</b>	InChI=1S/C14H14O2/c1-15-13-7-3-5-11(9-13)12-6-4-8-14(10-12)16-2/h3-10H,1-2H3
<b>InchiKey:</b>	UCHNVSDXSPIKRG-UHFFFAOYSA-N
<b>Formula:</b>	C14H14O2
<b>SMILES:</b>	<chem>COc1cccc(-c2cccc(OC)c2)c1</chem>
<b>Mol. weight [g/mol]:</b>	214.26
<b>CAS:</b>	6161-50-8

## Physical Properties

Property code	Value	Unit	Source
gf	62.56	kJ/mol	Joback Method
hf	-146.61	kJ/mol	Joback Method
hfus	21.70	kJ/mol	Joback Method
hvap	57.45	kJ/mol	Joback Method
log10ws	-4.31		Crippen Method
logp	3.371		Crippen Method
mcvol	172.340	ml/mol	McGowan Method
pc	2592.49	kPa	Joback Method
tb	601.20	K	NIST Webbook
tc	862.40	K	Joback Method
tf	369.88	K	Joback Method
vc	0.639	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	422.83	J/molxK	627.88	Joback Method
cpg	493.34	J/molxK	823.31	Joback Method
cpg	481.31	J/molxK	784.23	Joback Method
cpg	468.26	J/molxK	745.14	Joback Method
cpg	454.18	J/molxK	706.05	Joback Method
cpg	439.04	J/molxK	666.97	Joback Method

cpg	504.37	J/mol×K	862.40	Joback Method
dvisc	0.0001132	Paxs	627.88	Joback Method
dvisc	0.0001403	Paxs	584.88	Joback Method
dvisc	0.0001800	Paxs	541.88	Joback Method
dvisc	0.0002411	Paxs	498.88	Joback Method
dvisc	0.0003413	Paxs	455.88	Joback Method
dvisc	0.0005193	Paxs	412.88	Joback Method
dvisc	0.0008710	Paxs	369.88	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C6161508&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C6161508&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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