

# 4-Carbomethoxyhomocubane

|                      |                                                                            |
|----------------------|----------------------------------------------------------------------------|
| Inchi:               | InChI=1S/C11H12O2/c1-13-10(12)11-7-4-2-3-5(7)9(11)6(3)8(4)11/h3-9H,2H2,1H3 |
| InchiKey:            | VZNFAUDGDVVB-T-UHFFFAOYSA-N                                                |
| Formula:             | C11H12O2                                                                   |
| SMILES:              | COC(=O)C12C3C4CC5C3C1C5C42                                                 |
| Mol. weight [g/mol]: | 176.21                                                                     |
| CAS:                 | 40317-63-3                                                                 |

## Physical Properties

| Property code | Value           | Unit    | Source         |
|---------------|-----------------|---------|----------------|
| chl           | -5989.80 ± 5.00 | kJ/mol  | NIST Webbook   |
| gf            | 155.55          | kJ/mol  | Joback Method  |
| hf            | 26.00 ± 5.40    | kJ/mol  | NIST Webbook   |
| hfl           | -54.00 ± 5.00   | kJ/mol  | NIST Webbook   |
| hfus          | 27.22           | kJ/mol  | Joback Method  |
| hvap          | 80.00           | kJ/mol  | NIST Webbook   |
| log10ws       | -0.83           |         | Crippen Method |
| logp          | 0.917           |         | Crippen Method |
| mcvol         | 118.990         | ml/mol  | McGowan Method |
| pc            | 3149.09         | kPa     | Joback Method  |
| tb            | 521.68          | K       | Joback Method  |
| tc            | 726.36          | K       | Joback Method  |
| tf            | 407.89          | K       | Joback Method  |
| vc            | 0.503           | m3/kmol | Joback Method  |

## Temperature Dependent Properties

| Property code | Value  | Unit    | Temperature [K] | Source        |
|---------------|--------|---------|-----------------|---------------|
| cpg           | 386.58 | J/mol×K | 624.02          | Joback Method |
| cpg           | 398.14 | J/mol×K | 658.13          | Joback Method |
| cpg           | 408.95 | J/mol×K | 692.25          | Joback Method |
| cpg           | 345.11 | J/mol×K | 521.68          | Joback Method |
| cpg           | 360.29 | J/mol×K | 555.79          | Joback Method |
| cpg           | 374.04 | J/mol×K | 589.91          | Joback Method |
| cpg           | 419.22 | J/mol×K | 726.36          | Joback Method |

|       |              |         |        |              |
|-------|--------------|---------|--------|--------------|
| cpl   | 288.00       | J/mol×K | 298.00 | NIST Webbook |
| hvapt | 79.96        | kJ/mol  | 323.00 | NIST Webbook |
| hvapt | 80.00 ± 2.00 | kJ/mol  | 303.00 | NIST Webbook |
| hvapt | 80.00 ± 1.70 | kJ/mol  | 323.00 | NIST Webbook |

## Sources

|                        |                                                                                                                                               |
|------------------------|-----------------------------------------------------------------------------------------------------------------------------------------------|
| <b>McGowan Method:</b> | <a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>                         |
| <b>NIST Webbook:</b>   | <a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C40317633&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C40317633&amp;Units=SI</a> |
| <b>Crippen Method:</b> | <a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>                                     |
| <b>Crippen Method:</b> | <a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>                             |
| <b>Joback Method:</b>  | <a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>                                         |

## Legend

|                 |                                                           |
|-----------------|-----------------------------------------------------------|
| <b>chl:</b>     | Standard liquid enthalpy of combustion                    |
| <b>cpg:</b>     | Ideal gas heat capacity                                   |
| <b>cpl:</b>     | Liquid phase heat capacity                                |
| <b>gf:</b>      | Standard Gibbs free energy of formation                   |
| <b>hf:</b>      | Enthalpy of formation at standard conditions              |
| <b>hfl:</b>     | Liquid phase enthalpy of formation at standard conditions |
| <b>hfus:</b>    | Enthalpy of fusion at standard conditions                 |
| <b>hvap:</b>    | Enthalpy of vaporization at standard conditions           |
| <b>hvapt:</b>   | Enthalpy of vaporization at a given temperature           |
| <b>log10ws:</b> | Log10 of Water solubility in mol/l                        |
| <b>logp:</b>    | Octanol/Water partition coefficient                       |
| <b>mcvol:</b>   | McGowan's characteristic volume                           |
| <b>pc:</b>      | Critical Pressure                                         |
| <b>tb:</b>      | Normal Boiling Point Temperature                          |
| <b>tc:</b>      | Critical Temperature                                      |
| <b>tf:</b>      | Normal melting (fusion) point                             |
| <b>vc:</b>      | Critical Volume                                           |

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