

# 4-(Trifluoromethyl)benzyl alcohol

<b>Other names:</b>	p-Trifluoromethylbenzyl alcohol Benzenemethanol, 4-(trifluoromethyl)- 4-(Trifluoromethyl)benzylic alcohol
<b>Inchi:</b>	InChI=1S/C8H7F3O/c9-8(10,11)7-3-1-6(5-12)2-4-7/h1-4,12H,5H2
<b>InchiKey:</b>	MOOUWXDQAUZRG-UHFFFAOYSA-N
<b>Formula:</b>	C8H7F3O
<b>SMILES:</b>	OCc1ccc(C(F)(F)F)cc1
<b>Mol. weight [g/mol]:</b>	176.14
<b>CAS:</b>	349-95-1

## Physical Properties

Property code	Value	Unit	Source
gf	-599.15	kJ/mol	Joback Method
hf	-732.70	kJ/mol	Joback Method
hfus	16.04	kJ/mol	Joback Method
hvap	49.27	kJ/mol	Joback Method
log10ws	-2.72		Crippen Method
logp	2.198		Crippen Method
mcvol	111.000	ml/mol	McGowan Method
pc	3484.76	kPa	Joback Method
tb	500.86	K	Joback Method
tc	682.55	K	Joback Method
tf	283.87	K	Joback Method
vc	0.438	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	251.83	J/molxK	500.86	Joback Method
cpg	261.52	J/molxK	531.14	Joback Method
cpg	270.59	J/molxK	561.42	Joback Method
cpg	279.08	J/molxK	591.71	Joback Method
cpg	287.01	J/molxK	621.99	Joback Method
cpg	294.42	J/molxK	652.27	Joback Method

cpg

301.34

J/mol×K

682.55

Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	352.50 ± 0.50	K	0.50	NIST Webbook

## Sources

McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C349951&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C349951&amp;Units=SI</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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