

# Benzenamine, 2-methyl-5-nitro-

**Other names:** (2-Methyl-5-nitrophenyl)amine  
1-Methyl-2-amino-4-nitrobenzene  
2-Amino-4-nitrotoluene  
2-Methyl-5-nitro-benzeneamine  
2-Methyl-5-nitroaniline  
2-Methyl-5-nitrobenzenamine  
3-Nitro-6-methylaniline  
4-Nitro-2-aminotoluene  
5-Nitro-O-toluidine  
6-Methyl-3-nitroaniline  
Amarthol Fast Scarlet G Base  
Amarthol Fast Scarlet G Salt  
Azoene Fast Scarlet G Salt  
Azoene fast scarlet gc base  
Azoene fast scarlet gc salt  
Azofix scarlet g salt  
Azogene fast scarlet g  
Azoic diazo component 12, base  
C.I. 37105  
C.I. Azoic diazo component 12  
Conazoic Diazo AB  
Dainichi Fast Scarlet G Base  
Daito scarlet base g  
Devol scarlet b  
Devol scarlet g salt  
Diabase scarlet g  
Diazo fast scarlet g  
Fast Scarlet M 4NT Base  
Fast red sg base  
Fast scarlet base G  
Fast scarlet base j  
Fast scarlet g  
Fast scarlet g base  
Fast scarlet g salt  
Fast scarlet gc base  
Fast scarlet j salt  
Fast scarlet t base  
Hiltonil fast scarlet g base  
Hiltonil fast scarlet g salt  
Hiltonil fast scarlet gc base

Kayaku scarlet g base  
 Lake scarlet g base  
 Lithosol orange R base  
 Mitsui scarlet g base  
 NCI-C01843  
 Naphthanil scarlet g base  
 Naphtoelan fast scarlet g base  
 Naphtoelan fast scarlet g salt  
 PNOT  
 RCRA waste number U181  
 Scarlet base NSP  
 Scarlet base ciba II  
 Scarlet base irga II  
 Scarlet g base  
 Sugai fast scarlet g base  
 Symulon scarlet g base  
 o-Toluidine, 5-nitro-

**Inchi:** InChI=1S/C7H8N2O2/c1-5-2-3-6(9(10)11)4-7(5)8/h2-4H,8H2,1H3  
**InchiKey:** DSBIJCMXAIIKKKI-UHFFFAOYSA-N  
**Formula:** C7H8N2O2  
**SMILES:** Cc1ccc([N+](=O)[O-])cc1N  
**Mol. weight [g/mol]:** 152.15  
**CAS:** 99-55-8

## Physical Properties

Property code	Value	Unit	Source
chs	-3809.30 ± 1.30	kJ/mol	NIST Webbook
chs	-3811.70 ± 1.40	kJ/mol	NIST Webbook
gf	203.21	kJ/mol	Joback Method
hf	48.81	kJ/mol	Joback Method
hfus	23.71	kJ/mol	Joback Method
hvap	62.01	kJ/mol	Joback Method
log10ws	-2.23		Crippen Method
logp	1.485		Crippen Method
mcvol	113.130	ml/mol	McGowan Method
pc	4328.25	kPa	Joback Method
rinpola	272.81		NIST Webbook
rinpola	272.14		NIST Webbook
tb	620.57	K	Joback Method
tc	879.01	K	Joback Method

tf	446.98	K	Joback Method
vc	0.430	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	269.53	J/mol×K	620.57	Joback Method
cpg	280.15	J/mol×K	663.64	Joback Method
cpg	289.96	J/mol×K	706.72	Joback Method
cpg	299.00	J/mol×K	749.79	Joback Method
cpg	307.29	J/mol×K	792.87	Joback Method
cpg	314.89	J/mol×K	835.94	Joback Method
cpg	321.82	J/mol×K	879.01	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.25107e+01
Coeff. B	-3.79209e+03
Coeff. C	-8.72810e+01
Temperature range (K), min.	397.52
Temperature range (K), max.	614.02

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C99558&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C99558&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>

# Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h vap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>r in pol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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